

# 1 **Imaging air volume fraction in sea ice using non-** 2 **destructive X-ray tomography**

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## 14 15 **Abstract**

16 Although the presence of a gas phase in sea ice creates the potential for gas exchange with the  
17 atmosphere, the distribution of gas bubbles and transport of gases within the sea ice are still  
18 poorly understood. Currently no straightforward technique exists to measure the vertical  
19 distribution of air volume fraction in sea ice. Here, we present a new fast and non-destructive  
20 X-ray computed tomography technique to quantify the air volume fraction and produce  
21 separate images of air-volume inclusions in sea ice. The technique was performed on  
22 relatively thin (4 – 22 cm) sea ice collected from an experimental ice tank. While most of the  
23 internal layers showed air-volume fractions <2%, the ice-air interface (top 2 cm)  
24 systematically showed values up to 5%. We suggest that the air volume fraction is a function  
25 of both the bulk ice gas saturation factor and the brine volume fraction. We differentiate  
26 micro bubbles ( $\emptyset < 1$  mm), large bubbles ( $1 < \emptyset < 5$  mm) and macro bubbles ( $\emptyset > 5$ mm).  
27 While micro bubbles were the most abundant type of air inclusions, most of the air porosity  
28 observed resulted from the presence of large and macro bubbles. The ice texture (granular and  
29 columnar) as well as the permeability state of ice are important factors controlling the air  
30 volume fraction. The technique developed is suited for studies related to gas transport and  
31 bubble migration.

## 32 **1 Introduction**

33 Sea ice is a multi-phase system consisting of ice crystals, salt precipitates, brine, and  
34 gas bubbles (i.e. air inclusions). The abundance and morphology of brine and air inclusions  
35 are strongly dependent on the temperature and salinity of the sea ice (Cox and Weeks, 1983;  
36 Weeks and Ackley, 1986). Microscale studies of sea ice inclusions have in large part focused  
37 on the formation and morphology of brine inclusions (as pockets and/ or channels) (e.g.  
38 Bennington, 1967; Bock and Eicken, 2005; Cole and Shapiro, 1998; Cox and Weeks, 1975;  
39 Eicken et al., 2000; Eide and Martin, 1975; Galley et al., 2015a; Hunter et al., 2009; Notz and  
40 Worster, 2008). Inclusions in large part control the transfer of heat, salt, gases, and radiation  
41 between the ocean and atmosphere (Light et al., 2003). Brine and air inclusions in sea ice also  
42 affect the optical and electromagnetic properties of sea ice, and are often sites of biological  
43 activity (Fritsen et al., 1994, Krembs et al., 2000; Vancoppenolle et al., 2014).

44 Studies on the formation and morphology of gas inclusions and gas transport within  
45 sea ice are sparse. The air porosity quantitatively defined by the air volume fraction ( $V_a$  %)   
46 has generally been neglected in past work; it has long been assumed that gas species in sea ice  
47 were dissolved in brine and subject to the same processes as brine inclusions, and that the air  
48 volume fraction is minor compared to the brine volume fraction. Mushy layer theory, whose  
49 equations are now used as the physical basis for liquid exchange processes within sea ice  
50 neglect the presence of air inclusions (Feltham et al., 2006; Worster, 1992; Worster, 1997;  
51 Rees Jones and Worster, 2013). Omission of air inclusions in sea ice research propagates a  
52 lack of understanding of gas transport within sea ice, though studies in the last decade have  
53 revealed substantial CO<sub>2</sub> fluxes at the sea ice-atmosphere interface (Semiletov et al., 2004;  
54 Zemmelink et al., 2006; Delille et al., 20014; Geilfus et al., 2014, 2015; Normura et al., 2006;  
55 2010; 2013). It was also recently argued that a major part of the natural gases (oxygen,  
56 nitrogen and argon) as well as methane (CH<sub>4</sub>) reside in the gas phase inside bubbles in sea ice  
57 rather than dissolved in the brine (Zhou et al., 2013; Moreau et al., 2014; Crabeck et al.,  
58 2014a, b). Therefore the physical properties and processes of air inclusions in sea ice can  
59 potentially control the sea ice-atmosphere exchange of gases. The most important process  
60 leading to the formation of air inclusions from entrapped brine is brine volume reduction by  
61 freezing (Zhou et al., 2013; Crabeck et al., 2014a, b; Moreau et al., 2014). Increasing brine  
62 salinity during winter due to sea ice temperature reduction results in reduced gas solubility  
63 causing super-saturation (the *brine concentration effect*), which leads to bubble formation if

64 the sum of the partial pressures of all the dissolved gases is higher than the local hydrostatic  
65 pressure.

66 Previous studies of air inclusions morphology in sea ice were based on horizontal thin  
67 sections (e.g., Grenfell, 1983; Perovich and Gow, 1991, 1996; Light et al., 2003; Cole et al.,  
68 2004). Grenfell, (1983), Perovich and Gow, (1996) and Cole et al., (2004) highlighted that the  
69 columnar ice is usually depleted in air inclusions while top granular ice is described as bubbly  
70 with larger air inclusions. Grenfell (1983) measured bubble number distributions in small  
71 samples cut from first year sea ice, observing diameters ranging from 0.2 to 4 mm. Perovich  
72 and Gow (1996) reported mean bubble diameters ranging from 0.036 mm to 0.56 mm for 30  
73 cm thick pancake ice and mean diameter of 2.6 mm on a multi-year hummock. Light et al.,  
74 (2003) recorded 100 images from thin sections in transmitted light and reported bubble  
75 diameters between 0.008 mm to 0.14 mm in ice columnar ice that was 175 cm thick (Light et  
76 al., 2003).

77 Limitations of current methods have resulted in a lack of details on determination of  
78 air volume fraction. Those methods provide inadequate profiles of the vertical distribution of  
79 air inclusions in sea ice, especially in the context of ocean-sea ice-atmosphere exchange of  
80 gas. The sea ice air-volume fraction is most often determined empirically from bulk  
81 temperature, salinity and density measurements (after Cox and Weeks, 1983). However,  
82 small errors associated with sea ice density measurements result in large errors in the  
83 calculated air-volume fraction. Perovich and Gow (1996), and Light et al. (2003) used sea ice  
84 sections imaged using transmitted light to describe air inclusions within sea ice, given the  
85 caveats that undisturbed microstructure required careful thermal control, size may be limited,  
86 and the distinction between gas and brine can be ambiguous in transmitted images. While thin  
87 section studies are relevant to detail morphometric analysis of inclusions, profile of air  
88 volume fraction cannot be deduced from thin section analysis. Another approach is high  
89 resolution measurements of the total gas content along a vertical profile using techniques  
90 initially developed for continental ice cores (melting-refreezing and toepfer pump extraction  
91 or summing individual gases concentrations measured using gas chromatography (GC) (Tison  
92 et al., 2002)). These techniques however operate under vacuum, and therefore collect both the  
93 dissolved and gaseous phases. Also, this technique does not provide information on the  
94 morphology of the bubble content. A third approach used previously is to melt the ice sample  
95 in a gas tight container and quantify total gas volume (Rysgaard and Glud, 2004). A problem  
96 with this approach, however, is that gases equilibrate to a new bulk gas concentration  
97 depending on the salinity and temperature of the melting ice and hence do not represent the

98 actual gas volume at *in situ* conditions.

99 We propose a methodological advancement employing computed tomography (CT) X-  
100 ray imaging for measurement of air inclusions within sea ice. For many years CT X-ray has  
101 been widely used as a medical diagnostic tool. This non-invasive technique has largely  
102 contributed to the study of rock fractures and rock porosity, and has recently been applied to  
103 the sea ice field, advancing percolation theory for the brine system (Golden et al., 2007;  
104 Pringle et al., 2009, Obbard et al., 2009). Here we present high-resolution profiles of the  
105 distribution of air inclusions in sea ice, which are derived from CT X-ray images of whole ice  
106 cores at the sub-millimeter scale. A detailed statistical analysis of the air volume fraction in  
107 experimental sea ice is presented, as well as comparisons to the air volume equations of Cox  
108 and Weeks (1983) and measurement of total gas content. Throughout this work, we highlight  
109 the parameters and processes influencing the air porosity (air volume fraction,  $V_a$ ).

## 110 **2 Methods**

### 111 **2.1 Sea ice Environmental Research Facility (SERF)**

112 The Sea-ice Environmental Research Facility (SERF) at the University of Manitoba  
113 (Winnipeg, Canada) is an in-ground concrete pool with dimensions of 23.3 m (length) x 9.2 m  
114 (width) x 2.75 m (depth). It is filled each year with seawater formulated on site to closely  
115 replicate the chemistry of Arctic surface seawater (e.g. Hare et al., 2013, Geilfus et al., 2013,  
116 Rysgaard et al., 2014). In January 2013 an experiment was initiated from open water  
117 conditions, where sea ice was allowed to grow to 22 cm thick between January 13 and  
118 January 26. Ice cores were collected on January 14, 16 and 25 to measure bulk ice gas  
119 composition, temperature, salinity, and density, and for computed tomography (CT) X-ray  
120 imaging.

### 121 **2.2 Sea ice core: temperature, salinity and texture**

122 At least four ice cores were extracted on each sampling occasion using a Mark II core  
123 barrel with an internal diameter of 9 cm (Kovacs Ent., Lebanon NH, USA). One of the cores  
124 was destructively interrogated to measure an *in situ* ice temperature profile at a depth  
125 resolution of 2 cm using a calibrated probe (Testo 720, precision  $\pm 0.1^\circ\text{C}$ ) inserted into pre-  
126 drilled holes perpendicular to the ice core depth axis. The second ice core extracted was  
127 immediately cut into 2 cm slices which were stored in polyethylene buckets and left to melt  
128 close to  $0^\circ\text{C}$ . Bulk ice salinity of the melt of these 2-cm sections was derived from sample  
129 conductivity and temperature measured with an Orion Star Series WP-84TP conductivity

130 meter (precision  $\pm 0.1$ ) using the equations of Grasshoff et al. (1983). For sea ice gas content  
131 and CT X-ray imaging, the third and fourth cores were immediately wrapped in polyethylene  
132 bags and stored at  $-20^{\circ}\text{C}$  in the dark to ensure brine/gas immobilization and to inhibit  
133 biological processes (Eicken, 1991).

### 134 **2.3 Gas composition**

135 The bulk ice concentration of argon (Ar), oxygen ( $\text{O}_2$ ) and nitrogen ( $\text{N}_2$ ) expressed in  
136  $\mu\text{ mol L}^{-1}$  ice were analyzed using gas chromatography (GC). The dry-crushing technique as  
137 developed for gas measurements in continental ice (e.g. Raynaud et al., 1982) was used to  
138 extract the gas phase from the sea ice samples in a cold laboratory at  $-25^{\circ}\text{C}$ . Each ice core  
139 sample for gas composition was cut in 5-cm sections, and 60 g of each section were put into a  
140 vessel together with stainless steel beads which was evacuated to  $10^{-3}$  torr, and then fixed to  
141 an ice crusher (after Raynaud et al., 1982 and Stefels et al., 2012). The stainless steel beads  
142 impact the ice block during the shaking process, crushing it into a fine powder. After  
143 crushing, the vessel was kept in a cold ethanol bath ( $-50^{\circ}\text{C}$ ) and connected to a gas  
144 chromatograph (Trace GC) equipped with a thermal conductivity detector for concentration  
145 analyses (Skoog et al., 1997). We used Alphagaz<sup>TM</sup>2 He (Air Liquid –P0252) as the carrier  
146 gas and a 22 ml packed column (Mole Sieve 5A 80/100; 5 m x 1/8”). Gas collected included  
147 both gas bubbles in the ice and from the dissolved phase within the brine, which cannot be  
148 differentiated using this method. The total gas content ( $\text{ml L}^{-1}$  ice) was derived from the sum  
149 of the  $\text{O}_2$ ,  $\text{N}_2$  and Ar concentrations initially expressed in  $\mu\text{ mol L}^{-1}$  ice and applying the  
150 perfect gas law. Since both the cutting process and evacuation stage during the measurement  
151 process lead to potential gas lost, the total gas content measured is a minimum estimate of the  
152 true total gas content.

153 The saturation level of a gas affects bubble nucleation in brine inclusions and is  
154 therefore a crucial parameter determining gas flux at the ice-air interface. Theoretically,  
155 nucleation occurs when the sum of the partial pressures of dissolved gases is higher than the  
156 local hydrostatic pressure. We therefore compared (i) the gas concentrations profile measured  
157 in bulk ice;  $C_{\text{bulk ice}}$  to (ii) the theoretical inventory predicted by the solubility in brine at  
158 atmospheric saturation;  $C_{\text{Saturation}}$  (i.e. the maximum concentration of  $\text{O}_2$ ,  $\text{N}_2$  and Ar in the  
159 dissolved phase when the brine is not supersaturated (Carte, 1961; Lubetkin, 2003; Zhou et  
160 al., 2013)).  $C_{\text{Saturation}}$  is obtained by calculating the temperature and salinity-dependent  
161 solubility of  $\text{O}_2$ ,  $\text{N}_2$  and Ar in the brine (Garcia and Gordon, 1992; Hamme and Emerson,  
162 2004) and multiplying it by the relative brine volume (i.e. the brine volume fraction ( $V_b$ ), see

163 below) and expressed in  $\text{ml L}^{-1}$  of bulk ice. These relationships are valid for the range of  
164 temperature and salinity found in sea ice (Zhou et al., 2013). It is important to note that as  
165  $C_{\text{bulk ice}}$  is measured on 5 cm ice sections while  $C_{\text{saturation}}$  is computed using 2 cm sections, we  
166 can compute more than one  $C_{\text{saturation}}$  value for each  $C_{\text{bulk ice}}$ . The ratio between the gas  
167 concentration measured ( $C_{\text{bulk ice}}$ ) and the air concentration at equilibrium ( $C_{\text{saturation}}$ ) gives the  
168 saturation factor  $\text{SAT}_f$ . As a result, we present the mean  $\text{SAT}_f$  and its standard deviation for  
169 each  $C_{\text{bulk ice}}$  (5 cm) section. When a strong gradient of temperature, salinity, and therefore  
170 brine volume occurs in a  $C_{\text{bulk ice}}$  5 cm section, the standard deviation of  $\text{SAT}_f$  increases.

## 171 **2.4 Bulk ice density and Air volume Fraction**

172 To compute the brine volume fraction and the air volume fraction, the bulk ice density  
173 of 5-cm core sections was measured with the Mass-Volume technique in a cold lab ( $-20^\circ\text{C}$ )  
174 and the Cox and Weeks (1983) equations were then employed. Ice core sections were cut into  
175 cubes of  $5\text{ cm}^3$  and weighed precisely to determine their mass ( $M$ ). The dimensions of the  
176 sample were measured giving their volume ( $V$ ). The density of the ice ( $\rho_i$ ) calculated by:

$$177 \quad \rho_i = \frac{M}{V} \quad (1)$$

178 This common technique is easily applied, but there are several possible sources of  
179 errors: obtaining a dimensionally perfect ice sample is difficult, and inaccuracies in the  
180 measurement of the sample dimensions lead to volume error (Timco and Frederking, 1995).  
181 To limit error induced by imperfect sample dimensions, we used a precision diamond wire  
182 saw. The length of each edge (the number of edges per cube = 12) was found to deviate from  
183 5 cm by  $\pm 0.07$  cm on average (total number of edges measured = 96; 8 ice cubes) yielding an  
184 average precision for ice density of  $\pm 4.4\%$ , as a result of the cutting process. While deviation  
185 of 0.7 mm on the dimension of the ice cube has little effect on the precision of the density and  
186 of the calculated brine volume fraction (relative standard error  $< 5\%$ ), it produces relative  
187 standard errors as high as 163% on the air volume fraction computed using the equations of  
188 Cox and Weeks (1983) (Table 1).

## 189 **2.5 Liquid porosity: brine volume fraction**

190 The brine volume was calculated according to Cox and Weeks (1983) using in situ  
191 temperature, bulk ice salinity, and bulk ice density measurements from the cores. Brine  
192 salinity ( $S_b$ ) was calculated using in situ sea ice temperatures and the freezing point of  
193 seawater (UNESCO, 1978). The brine volume fraction, ( $V_b$ , expressed in %), was calculated  
194 from the ratio of brine volume and bulk sea ice volume ( $b/V$ ). In previous works, sea ice air

195 volume fraction is ubiquitously neglected, so historically, sea ice porosity refers solely to the  
196 brine volume fraction. In the context of this paper, the terms brine inclusions and brine  
197 volume fraction refers to liquid porosity. The permeability threshold of  $V_b = 5\%$  following  
198 Golden et al (1998, 2007) defines permeable and impermeable columnar sea ice.

## 199 **2.6 Ice texture:**

200 To describe the ice crystal texture, horizontal thin sections of maximum 10 cm length  
201 were produced in a cold lab at  $-20^\circ\text{C}$  using the standard microtome (Leica SM2400)  
202 procedure described by Langway (1958) and Tison et al. (2008). Images of these backlit  
203 horizontal thin sections were taken in the cold lab between crossed polarizing sheets with a  
204 camera (Nikon Coolpix S200).

## 205 **2.7 Air porosity: air volume fraction by CT X-ray imaging:**

### 206 **2.7.1 General principle**

207 CT scanning is a non-destructive radiographic approach to examine materials by  
208 creating a three-dimensional image of density contrasts. Ice cores were imaged using a third  
209 generation Siemens Somatom Volume Access sliding gantry medical CT-Scanner (Siemens  
210 SOMATOM Definition AS+ 128) at the Institut National de la Recherche Scientifique (INRS-  
211 ETE). The ice cores were stored at  $-20^\circ\text{C}$  and scanned at room temperature. The scan duration  
212 was less than 15 seconds. Including transport of the ice samples from the storage freezer to  
213 the CT instrument was less than 75 seconds. We therefore assume that no temperature change  
214 in the core occurred. Data was acquired in spiral mode with a pitch factor of 0.6; the X-ray  
215 source was set at 120kV and 150 mAs. These configurations produced 1152 projections for  
216 each reconstructed axial slice. The image size is limited by the manufacturer to 512 x 512  
217 pixels, so the pixel resolution is defined by the chosen field of view (FOV). The smallest  
218 selectable FOV is 50 x 50mm providing a pixel resolution of 0.0977 mm in the transverse  
219 plane. This FOV is too small to contain the whole core in one image; so four reconstructions  
220 of each core were produced and concatenated together using Matlab. The Siemens SAFIRE  
221 (Sinogram Affirmed Iterative Reconstruction) reconstruction algorithm was used (three  
222 iterations). The convolution kernel is J70h, a medium-sharp filter. The result of the  
223 concatenation is an image size of 1024 x 1024 pixels with a FOV of 100 mm x 100 mm and a  
224 pixel resolution of 0.097 mm (x-y) and a slice thickness (z) of 0.6 mm. By scanning a core  
225 from top to bottom, a three-dimensional "stack" of images was produced by compiling  
226 individual transverse slices and longitudinal slices (Fig. 1a) yielding  $0.097 \times 0.097 \times 0.6$  mm

227 (0.0056 mm<sup>3</sup>) voxel volumes within square (1024 x 1024 voxel) images.

228 Hounsfield (1973) and Knoll (1989) describe the X-ray technique in detail. The  
229 Hounsfield Unit (HU) value for each voxel corresponds to linear X-ray attenuation (Duliu,  
230 1999), where higher density and higher atomic numbers result in greater X-ray attenuation.  
231 Ice core density was calculated in terms of tomographical intensity (TI) (in Hounsfield Units  
232 for each voxel):

$$233 \quad TI = \left( \frac{\mu}{\mu_w} - 1 \right) * 1000 \quad (2)$$

234 Where  $\mu$  is the linear absorption coefficient of the bulk core, and  $\mu_w$  is the linear absorption  
235 coefficient of water.  $\mu$  is a function of the radiation energy and the atomic number of the  
236 core component crossed by the beam and varies in relation to the density of the material.  
237 Resulting images are represented in grey scale where darker tones indicate lower density  
238 material (e.g. air) (Fig.1a-c). Density measurements were made relative to freshwater and  
239 expressed in TI where water = 0 HU and air = -1000 HU. We observed Hu unit from -1024 to  
240 +616, +499 and +766 on January 14, 16 and 25 respectively. Positive values are related to  
241 brine and to a minor extend precipitated salt (e.g. 60 HU < TI < 766 HU) and slightly negative  
242 values are related to ice (e.g. TI = -84 HU, Kawamura et al., (1988)). We estimate the  
243 tomographic intensity of pure ice crystals using the mode of the histogram (Fig. 1d, TI = -74  
244 HU).

### 245 **2.7.2 Processing X-ray Images and analysis**

246 The process of pixel selection to create binary images of air inclusions, therefore  
247 defining the air volume fraction (air porosity,  $V_a$ ) in the CT imagery was performed by means  
248 of thresholding following the determination of a region-of-interest (ROI) created by the  
249 removal of all pixels not belonging to the core sample (i.e. the sample container, disturbance  
250 of the core edges by coring and/or storage, the supporting bench, and surrounding air from  
251 each slice). The images were individually examined and ambiguous pixels around the sea ice  
252 core sample were also removed. The bottom slices of each imaged ice core were removed due  
253 to observed brine drainage resulting from the coring method. A three-dimensional orthoslice  
254 view as well as two transversal slices of the ice core sample extracted on January 16 are  
255 shown in Fig. 1. The bottom of the ice core from which brine drainage occurred during core  
256 extraction can be clearly differentiated from the rest of the core sample unaffected by brine  
257 loss on both the orthoslice and transversal slice views (Fig. 1a-c). CTan and ImageJ software

258 were used to quantitatively measure morphometric characteristics of binary (black and white)  
259 images.

260 Determination of the most applicable threshold is therefore of the utmost importance  
261 here, as in all image classifications in the multitude of fields that employ the technique. Three  
262 approaches are typical for determining an optimal threshold; manual threshold selection based  
263 on the human visual system, automated threshold selection based on image data, usually  
264 employing the image histogram, and a threshold based on a mixture model approach.

265 There are many automated segmentation techniques described in the literature  
266 exceeds. In this study segmentation algorithms representing a selection of established  
267 thresholding techniques chosen on the basis that they (i) suited a unimodal histogram (Fig.  
268 1d), and they (ii) showed potential for automated characterization of pore space in  
269 geomaterials. Global thresholding specifically was selected on the basis of comparative  
270 reviews by Sezgin and Sankur, (2004) and Iassonov et al. (2009). Global thresholding may be  
271 divided into several subcategories depending on the applied approach. These subcategories in  
272 include those based on signal entropy considerations (Shannon and Weaver, 1948; Pal and  
273 Pal, 1989; Pal, 1996) to separate background and foreground voxels, including EN-Kapur and  
274 EN-Yen (Kapur et al., 1985; Yen et al., 1995). There exist global thresholding methods that  
275 analyze histogram shape (HS), including HS-Zack and HS-Tsai (Zack et al., 1977; Tsai,  
276 1995). Finally, segmentation may be accomplished by clustering (CL) methods, which  
277 separate background (i.e. ice) and foreground voxels (i.e. air) by approximating the histogram  
278 with a combination of two or more statistical distributions, including CL-Otsu and CL-Ridler  
279 (Ridler and Calvard, 1978; Otsu, 1979).

280 Each segmentation method was tested on the three core image sets (633 total image  
281 slices), as well as on selected parts of each image set to insure that the algorithm response was  
282 stable. The results of each segmentation method were visually evaluated by comparing the  
283 raw and segmented images (Fig. 2) and by computing linear profiles of Hu-value (Fig. 3)  
284 through cross-sectional images and examining them visually to determine the efficacy of  
285 various thresholds in identifying air inclusions.

286 Analysis of variance (ANOVA) demonstrated significant ( $p < 0.005$ ) differences  
287 between in the thresholds produced by the interrogated segmentation methods. The EN-Yen  
288 (max-entropy) algorithm produced a relatively high threshold (TI = -200 HU) but also  
289 introduced noise and speckle in the image; 66% of 1180 total inclusions detected by the EN-  
290 Yen threshold were the size of a single pixel (Fig. 2b, EN-Yen). The HS-Zack (Zack et al.,  
291 1977; Rosin, 2001) method produced a low threshold (TI= -569 HU) that (i) did not detect

292 small bubbles, and (ii) underestimated the size of bubbles detected (Fig. 2c. HS-Zack.). The  
 293 segmentation threshold produced by the CL-Ridler (TI = -370 HU), method accurately  
 294 identified bubbles in all images, including detecting very small bubbles, without introducing  
 295 speckle in the segmented image (Fig. 2d. CL-Ridler).

296 Manual segmentation thresholds were defined by inspecting a variety of different  
 297 bubbles in different slices, (e.g. Fig. 3). Figure 3 indicates that visual thresholds were  
 298 subjective; the pixel scale actually makes visual bubble delineation more ambiguous. Bubble  
 299 number 2 (Fig. 3) is best delineated by TI = -453 HU, while bubble number 3 (Fig. 3) is best  
 300 visually delineated by TI = -373 HU. In the context of the variety of bubble morphology and  
 301 the differences in columnar and granular sea ice in the 633 transverse image slices, the visual  
 302 segmentation threshold was set at TI = -453 HU.

303 Finally, the tomographic intensity of “mixed pixels” which appear as varying shades  
 304 of grey is dependent of the proportion of air ( $V_{air}$ ), ice ( $V_{ice}$ ) an brine ( $V_{brine}$ ) in the pixel and  
 305 the proportions of the tomographic intensities of those constituents (air ( $Hu_{air}$ ), ice ( $Hu_{ice}$ ) and  
 306 brine ( $Hu_{brine}$ )) in the following mixture model:

307

$$308 \quad Hu = (V_{air} \times Hu_{air}) + (V_{ice} \times Hu_{ice}) + (V_{brine} \times Hu_{brine}) \quad (3)$$

$$309 \quad 1 = V_{air} + V_{ice} + V_{brine} \quad (4)$$

310

311 The TI of pure ice crystals is determined using the mode of the histogram containing  
 312 all the data from each core ( $TI_{pure\ ice} = -74$  HU). Brine TI values ranged from 60 to 500  
 313 depending on brine salinity; we selected the middle point of that range ( $TI_{brine} = 200$  HU).  
 314 Finally  $TI_{air} = -1000$  HU. According to the mixture model (equation (3)), any pixel  $TI \geq -400$   
 315 HU contains at least 50% air and is therefore selected as part of an air inclusion (Fig. 2e,  
 316 Table 2).

317 The CL-Ridler (TI = -370 HU) threshold, as well as the manual threshold (TI = -453  
 318 HU) and the mixture model threshold, which selects pixel containing 50% air (TI = -400 HU)  
 319 were used to compute the air volume fraction for each of the transverse slices in each of the  
 320 three ice cores imaged. Analysis of variance (ANOVA) demonstrated no significant  
 321 difference ( $p < 0.005$ ) between the air volume fraction computed using the mixture model  
 322 threshold and the air volume fraction computed using the visually defined threshold (TI = -  
 323 453 HU) or the most applicable automated threshold (CL-Ridler, TI = -370 HU). The CL-  
 324 Ridler and the visual threshold produced mean  $V_a$  values that were statistically different ( $p <$   
 325 0.005)

326 Hereafter, the air volume fraction is presented as the mean of the air volume fraction  
327 results computed using the three selected thresholds. The potential range of the  $V_a$  from each  
328 of the three methods is represented by the standard deviation of this mean. We are able to  
329 compute the  $V_a$  in granular layer within a potential range of  $\pm 16\%$  and in columnar sea ice  
330 within a potential range  $\pm 43\%$ .

331 Our method endeavours to meet the challenge of CT X-ray image threshold selection  
332 in porous materials while lacking knowledge of the optimal segmentation result. Selecting the  
333 most applicable threshold is imperfect because the resolution of the CT-imaged used will  
334 almost always be insufficient to resolve every object of interest (in this case air inclusions in  
335 sea ice). When the object of interest is smaller than the spatial resolution of the imager, it  
336 appears as a mixed pixel, where the tomographic intensity is a combination of air and the  
337 presence of ice and/or brine, resulting in voxel TI's than that of pure ice by some amount. In  
338 this way delineation of an object using TI thresholds is complicated by the TI's of adjacent  
339 pixels/materials. If an air bubble (TI = -1000 HU) is adjacent to both ice (TI = -74 HU,  
340 Fig.1d) and brine (TI = 200 HU) the pixel(s) those respective boundaries will be roughly -537  
341 on the ice-side and -400 on the brine-side. In granular sea ice where bubble are sufficiently  
342 large to be resolve by Ct-scan, mixed pixel concerns mainly the edges of the large and macro  
343 bubbles, while in columnar sea ice where bubble are small compared to the pixel size, most of  
344 the bubbles appeared as mixed pixels as they include both air and background (ice/brines).  
345 We are able to compute the  $V_a$  in granular layer within a potential range of  $\pm 16\%$  and in  
346 columnar sea ice within a potential range  $\pm 43\%$ . The morphology of air inclusions is  
347 characterized quantitatively using their diameters ( $\emptyset$ , mm) in the transverse (x-y) plane.  
348 While it is ambiguous to report exact diameter from mixed pixel, we classified bubble  
349 diameters into three categories at a millimeter scale: micro bubbles ( $\emptyset < 1$  mm), large bubbles  
350 ( $1\text{mm} < \emptyset < 5\text{mm}$ ), and macro bubbles ( $\emptyset > 5\text{mm}$ ).

351  $V_a$  must be clearly differentiated from the GC-derived bulk ice total gas content (in ml  
352  $L^{-1}$  ice) which refers to the amount of  $O_2$ ,  $N_2$ , and Ar, both (i) in dissolved phase in brine and  
353 (ii) in the gas phase in bubbles measured in 5-cm depth increments. In this work we use the  
354 terms “bubbles” and “air inclusions” interchangeably to denote gas phase inclusions in sea  
355 ice.

## 356 **3 Results**

### 357 **3.1 Environmental conditions**

358 At the Sea ice Environmental Research Facility (SERF), the ambient air temperature varied  
359 between  $-5^{\circ}\text{C}$  and  $-32^{\circ}\text{C}$  through the experiment from January 13-26. The average air  
360 temperature for the period was  $-22^{\circ}\text{C}$ . Three main snowfall events occurred during the  
361 experiment. Snowfall on 14 to 15 January covered the sea ice surface with 1 cm of snow.  
362 Snowfall from January 19 to 23 deposited 6-9 cm of snow over the entire pool. On the  
363 morning of January 23, the snow was manually cleared off the ice surface to investigate the  
364 insulating effect of snow on the ice temperature and ikaite precipitation (see Rysgaard et al.,  
365 2014). Finally, from January 24 to January 27, 8 cm of snow covered the entire pool until the  
366 end of the experiment on January 30. Surface elevation from Lidar data (not shown), indicate  
367 that ice surface did not move appreciably in the vertical for the duration of the experiment  
368 even as a result of snow removal. Periodically the sea ice froze to the side of the pool  
369 resulting in a hydrostatic pressure head in the seawater below, causing episodic percolation of  
370 seawater at the freezing point upwards through the sea ice volume resulting in wet snow ice  
371 and slush at the sea ice surface on 16 Jan. A slush layer (up to 3 cm thick) was also observed  
372 at the snow base on 20 Jan. This episodic hydrostatic pressure head and resultant upward  
373 percolation of seawater through the sea ice caused the granular layer of the sea ice volume to  
374 thicken over time, likely by the formation of snow ice layers as the slush layer froze. On  
375 January 14, the granular layer was 0.7 cm frazil ice, on January 16, the granular layer  
376 thickened to 1.7 cm (consisting of the initial 0.7 cm of frazil and 1 cm of snow ice). On  
377 January 25, the granular layer had thickened again to 4 cm consisting of the initial granular  
378 sea ice layer of 14 Jan, the snow ice layer of 16 Jan and an additional 2.3 cm thick snow ice  
379 layer (Fig. 4). Below the granular ice layer, the sea ice crystal texture transitioned nearly  
380 immediately to columnar ice on all three dates (Fig. 4).

### 381 **3.2 Temperatures, salinity, brine volume fraction and bulk ice density**

382 Sea ice temperature, bulk salinity, brine volume and bulk ice density profiles for cores  
383 sampled on January 14 (4 cm thick), January 16, (8 cm thick) and January 25 (22 cm thick)  
384 are shown in Fig. 5 and Table 3.

385 On January 14, the bulk salinity profile was approximately linear, and evolved to a  
386 more a C-shaped profile on January 16 and 25 as the granular top layer remained saline and  
387 the top of the columnar layer desalinated through the experiment (Fig. 5). Calculated brine

388 volume ( $V_b$ ) profiles were similar in shape to the salinity profiles with minimum  $V_b$  occurring  
389 in the middle of the columnar ice layer on 16 and 25 Jan (Fig. 5). According to Golden et al.,  
390 (1998, 2007) the permeability threshold for columnar sea ice of 5%  $V_b$  indicates the whole ice  
391 volume on 14 Jan and near the bottom parts of the columnar ice layer on 16 and 25 Jan were  
392 permeable to liquid.

393 Bulk ice densities ranged from  $0.84 \text{ g cm}^{-3} \pm 0.020 \text{ g cm}^{-3}$  to  $0.92 \text{ g cm}^{-3} \pm 0.023 \text{ g cm}^{-3}$ .  
394 The lowest densities were systematically found at the surface of the ice cover (Fig. 5).

### 395 **3.3 Bulk ice total gas content**

396 The total gas content in the sea ice volume increased from its minimum in the bottom  
397 permeable columnar layer to its maximum in the top granular layer on 14, 16 and 25 Jan. The  
398 total gas content in the sea ice volume also increased over time (Fig. 6). The total gas content  
399 of the permeable columnar bottom of each of the ice cores (and the entire core on 14 January)  
400 were close to the concentration at saturation with respect to calculated theoretical atmospheric  
401 gas concentrations, leading to saturation factor ranging from 0.8 to 1.2. This will be referred  
402 to as “subsaturated” ( $\text{SAT}_f \leq 1.2$ ). On the contrary, the total gas content of the impermeable  
403 columnar layers and the granular surface layers of the sea ice were largely greater than the  
404 concentration at saturation leading to saturation factor ranging from 9.5 to 16. These will be  
405 referred to as supersaturated ( $\text{SAT}_f > 1.2$ ).

### 406 **3.4 Air Porosity**

#### 407 **3.4.1 Air volume fraction ( $V_a$ ) derived from CT X-ray Image Analysis**

408 For each of the three dates sampled, the air volume fraction increased from bottom  
409 columnar ice layer to the granular surface ice layer and the CT-derived air volume fraction in  
410 the sea ice increased overall from 14 to 25 Jan (Fig. 4 and Fig. 7a-c) in the same way as was  
411 shown by the total gas content analysis (Fig. 6).

412 In columnar ice, we distinguish permeable ( $V_b > 5\%$ ) and subsaturated ( $\text{SAT}_f \leq 1.2$ )  
413 ice near the bottom (Fig. 7a and b, shaded blue) from the impermeable ( $V_b < 5\%$ ) and  
414 supersaturated ( $\text{SAT}_f > 1.2$ ) ice in the middle sea ice layers (Fig. 7a and b, white area). In the  
415 permeable subsaturated bottom sea ice,  $V_a < 1\%$ , for each sampling date (Table 3). In the  
416 intermediate supersaturated impermeable columnar layer, the air volume fraction was  
417 typically under 2% and increased from 14 to 25 Jan. At the transition between the  
418 impermeable columnar ice and the permeable columnar ice on 16 and 25 Jan (Fig. 7c, shaded  
419 pink), we observed a slight increase in  $V_a$ .

420 On all three dates, the maximum air volume fraction occurred in the granular layers  
421 nearest the atmosphere interface increasing from the base of the granular layer (Fig. 7c and d).  
422 As the granular ice layer thickened by snow ice formation from 0.7 cm to 4 cm,  $V_a$  in these  
423 layers increased (Fig. 7c and d). In the granular layers, the brine volume exceeded 5% (Fig. 5)  
424 and the saturation factor  $> 9$  on 16 and 25 Jan (Fig. 6).

### 425 **3.4.2 Air inclusion morphology**

426 The morphology of air inclusions is characterized quantitatively using their diameters  
427 ( $\emptyset$ , mm) in the transverse (x-y) plane (Fig. 8). We classified CT-derived bubble diameters  
428 into three categories: micro bubbles ( $\emptyset < 1$  mm), large bubbles ( $1 \text{ mm} < \emptyset < 5\text{mm}$ ), and  
429 macro bubbles ( $\emptyset > 5$  mm) (see for e.g. Fig. 8). Bubbles smaller than the pixel size (0.0097  
430 mm in the transverse plane) represented less than 10% of the bubble population in any type of  
431 ice except for the impermeable supersaturated columnar layer on 16 Jan (Fig. 9a). Most of the  
432 bubbles had diameters  $\leq 1$  mm; bubbles of this size were common at every depth in every ice  
433 type interrogated by the CT imager (Fig. 9a and b). Due to the non-destructive nature of the  
434 CT X-ray method we were able to observe larger bubbles with diameters as large as 18.3 mm  
435 in granular sea ice (Fig. 9b).

436 For each ice core, the bubble size increased from the bottom columnar layer, to the top  
437 granular layer and increased over time in the same way as observed by the total gas content  
438 measured using the GC method (Fig. 9b). The bottom permeable subsaturated columnar ice  
439 contained almost exclusively micro bubbles on all three dates (Fig. 9a and b). Large bubbles  
440 occurred more frequently in the intermediate impermeable supersaturated columnar layer than  
441 in the bottom permeable and subsaturated columnar layer. Macro bubbles ( $\emptyset > 5$  mm) were  
442 exclusively found close to the ice-atmosphere interface in the snow ice layer (Fig. 9b).

## 443 **4 Discussion**

### 444 **4.1 Computed tomography X-ray imaging as a non destructive method to** 445 **compute the sea ice air volume fraction**

446 By using computed tomography X-ray imaging with a voxel size of  $0.0056 \text{ mm}^3$  we  
447 provide high-resolution profiles of the vertical distribution of air inclusions in sea ice, from  
448 which the sea ice air volume fraction are computed every 0.6 mm. Taking into account the  
449 relative errors of  $V_a \pm 16\%$  in granular layer and  $V_a \pm 43\%$  in columnar ice, results of image  
450 analysis indicated that the air volume was  $< 1\%$  in most of the columnar ice, but

451 systematically reached 5% in the granular/snow ice top layer where an air volume fraction as  
452 high as 19% was observed.

453 CT X-ray images (of porous materials in particular) are of such high resolution (in this  
454 case voxel = 0.0056 mm<sup>3</sup>), use such large sample volumes and are so quick that traditional  
455 methodology can hardly be used to validate the imaged data at the same resolution.  
456 Nevertheless, we compared our CT-derived  $V_a$  results to air volume fraction computed based  
457 on density measurements (Cox and Weeks, 1983) (Fig. 10a). The density (M/V) derived air  
458 volume profiles were always larger (Fig. 10a) but both methods derive large difference  
459 between granular and columnar air porosity and showed similar trends (Fig. 10b). The  
460 precision in density-derived  $V_a$  is very low ( $\pm 163\%$ ), compared with the relative standard  
461 deviation from CT- derived volume fraction of  $V_a \pm 16\%$  in granular layer and  $V_a \pm 43\%$  in  
462 columnar sea ice. The CT-derived air volume fraction also compared to the bulk ice total gas  
463 content (ml L<sup>-1</sup> ice) data derived from the GC analysis (using 60 g samples from 5-cm thick  
464 sections) (Fig. 10c). The vertical gradients in the two datasets increased similarly from the ice  
465 bottom to the sea ice surface and both the total gas content (ml L<sup>-1</sup> ice) and the CT-derived air  
466 volume fraction increased as the ice thickened over time.

467 Correlation between CT-derived air volume fraction and the bulk ice gas content (ml  
468 L<sup>-1</sup> ice; Fig. 10c) is not straightforward due in part to methodological constraints. The data  
469 compared well in columnar sea ice while in granular layer, we observed large deviations  
470 (Fig.10c). In granular ice, the total gas content was much lower than the CT-derived air  
471 volume fraction. CT image voxels are 0.0056 mm<sup>3</sup>, whereas the bulk ice total gas content was  
472 measured on 5-cm sections. Those 5-cm thick sections did not always resolve the changes in  
473 ice type. Within the 5-cm sections, gas might span a large range of concentration as does the  
474 air volume fraction in granular sea ice (Fig.10b, error bar); therefore if thinner sections had  
475 been analyzed for the total gas content (ml L<sup>-1</sup> ice) the values obtained might have been  
476 higher in the top part of the ice core, similar to the CT image data. Moreover, the  
477 measurement of the total gas content is a destructive sampling process, involving multiple  
478 steps in which the gas could potentially leave the ice. During the cutting process, some  
479 bubbles are inevitably cut in half, so part of the gas is lost. This is further complicated by the  
480 fact that the probability of cutting a large bubble (with high gas content) is greater than for a  
481 small bubble with low gas content. Potential gas loss could also happen during the evacuation  
482 phase of the measurement of the total gas content (i.e.. section 2.3). For all these reasons, the  
483 total gas content (ml L<sup>-1</sup> ice) is likely to be particularly underestimated in the granular surface  
484 ice due the analytical procedure. The data however agree well in columnar sea ice. The bulk

485 ice total gas content measured by gas chromatography includes both gas dissolved in brine  
486 and gas trapped in bubbles, while CT-derived air volume fraction results only account for gas  
487 trapped in bubbles. Then, we expect the total gas content values to be slightly higher than the  
488 CT-derived air volume fraction due to the dissolved contribution. In reality, the total gas  
489 content appears to be slightly lower (Figure 10c). It suggests that either the CT-derived air  
490 volume fraction is slightly overestimated in columnar sea-ice due to the thresholding process  
491 or that the total bulk gas is slightly underestimated due to gas loss during the cutting and  
492 evacuation phase of the measurement process. Finally, the bulk ice total gas content was  
493 measured on different ice cores from those used for the CT measurements, which may have  
494 introduced some spatial variability.

495           The temperature during storage finally potentially influences our computation.  
496 Storing sea ice at  $-20^{\circ}\text{C}$  alters the sea ice microstructure and its inclusions (e.g. Cox and  
497 Weeks, 1986). Light et al., (2003) proceeded to a cooling sequence ( $-2^{\circ}\text{C}$  to  $-25^{\circ}\text{C}$ ) and a  
498 warming sequence ( $-25^{\circ}\text{C}$  to  $-2^{\circ}\text{C}$ ) on ice thin sections. According to their results cooling sea  
499 ice caused inclusions to shrink in size including the disappearance of the smallest air  
500 inclusions, while warming increased the size of existing air inclusions without forming new  
501 bubbles. Considering that the smallest bubbles could have disappeared and some have shrunk  
502 in size, our computed air volume fraction should be considered as a minimum estimate of the  
503 true air volume fraction. In the absence of a method that preserves the natural temperature  
504 gradient within sea ice immediately and without change upon extraction, ex situ analysis of  
505 sea ice samples after storage at low temperatures is an established protocol.

506           Although microstructural analysis of sea ice may produce reliable morphological results  
507 for air inclusions, thin sections only represent a small subsample of the ice core, are time  
508 consuming, can be operator-dependent, and the area and thickness of a thin section limit these  
509 studies to the measurement of intact bubbles within a thin section. Density-derived air volume  
510 fraction results from the Mass-Volume technique generally have large errors and very low  
511 vertical resolution because they require large core subsample volumes (e.g.  $5\text{ cm}^3$ ). On the  
512 contrary, CT X-ray imaging clearly distinguishes between air inclusions and the ice matrix  
513 providing high-resolution sub-millimeter profiles of the air volume vertical distribution with a  
514 better precision linked almost entirely to the segmentation process and the resolution of the  
515 scanner. X-ray tomography allows: (1) fast visualisation of the air inclusions, especially when  
516 compared to transmitted images; (2) the ability to increase the size of the dataset compared to  
517 thin section microstructural analysis by imaging the whole core. Future work should involve  
518 micro CT –X-ray with a voxel resolution of an order of magnitude higher than the present one

519 in order to detect the small bubbles in columnar sea ice as well as research on the effect of  
520 temperature changes on sea ice gas inclusions.

#### 521 **4.2 Size of the air inclusion (i.e. bubbles): Micro, large and macro air porosity**

522 While large and macro bubbles account for less than 17%, 24% and 27% of the bubble  
523 population observed for January 14, 16 and 25 respectively (Fig. 11a), the large and macro  
524 bubbles contribute systematically to more than 50% of the total air volume fraction produced  
525 (Fig. 11a). Even in bottom columnar ice where large bubbles represent only 10% of the  
526 bubble population (Fig. 11d), they contributed to 40 % and 22 % of the air porosity of bottom  
527 columnar ice on January 14 and 25 respectively (Fig. 11d). For each ice type (Fig. 11,  
528 granular (b.), columnar impermeable (c.) and columnar permeable (d.)), it is clear that the  
529 largest bubbles contribute most to the air porosity (Fig. 11, Table 4), which is not surprising  
530 as the latter depends on air bubble size cubed. However air porosity in the permeable  
531 columnar layer where the proportion of large bubbles decreased (Fig. 12) seems largely to be  
532 controlled by the amount of bubbles (i.e. bubble density number). Increasing the number of  
533 bubbles produces also a linear increase in the air volume fraction (Fig. 12) in columnar ice.

534 For the air volume fraction to increase above 3% (e.g. Fig. 12 in the granular layer),  
535 the presence of large and macro bubbles are required (Fig. 12).

536 Large bubbles were more prevalent when brine volume increased (Fig. 13a, red and grey  
537 circles). Light et al. (2003) observed that bubbles were contained within brine and concluded  
538 that bubble size was limited by the size of the brine inclusion in which they resided. In several  
539 slices, we observed lighter pixels around air inclusions indicating these bubbles likely formed  
540 in a brine pocket. The CT-scanner used here cannot unambiguously identify these pixels as  
541 brine inclusions. To visualize both air and brine inclusions in the same images, finer  
542 resolution with respect to sample density and finer spatial resolution are required. For  
543 example, Obbard et al. (2009) showed that micro-X-ray computed tomography with a higher  
544 voxel resolution of one order of magnitude is suitable for visualization of brine and air  
545 inclusions.

#### 546 **4.3 Mechanism for gas incorporation and bubbles development**

547 In our sea ice samples, the top granular layers are supersaturated, have large air  
548 volume fractions ( $V_a > 5\%$ ) and contain macro bubbles. The impermeable columnar layers are  
549 supersaturated as well but contain lower air volume fractions ( $V_a < 2\%$ ) and contain micro to  
550 large bubbles. The bottom permeable columnar layers are subsaturated, contain air volume  
551 fractions  $< 1\%$ , and contain almost exclusively micro bubbles (see summary Table 5).

552 In the multiphase sea ice system, the ratio between dissolved gas and bubbles should  
553 depend on the bulk ice gas saturation state. In a closed system, when bubble nucleation is  
554 exclusively solubility driven, we expect the air volume fraction to be a function of the  
555 saturation factor, which would lead to subsaturated sea ice being bubble-free, and high air  
556 volume fractions in supersaturated sea ice. However, the observed relationship between air  
557 volume fraction and saturation factor is not straightforward (Fig. 13b) and highlights  
558 difference between the type of ice (i) bottom permeable columnar ice, (ii) intermediate  
559 impermeable columnar ice and (iii) top granular ice.

#### 560 **4.3.1 Bottom permeable columnar ice**

561 Within the permeable subsaturated columnar layer near the sea ice bottom, the air  
562 volume fraction is lower than 1 % due to the subsaturated state of the ice, and independent of  
563 the brine volume fraction (Fig. 13a and b, blue circles). As long as the brine is able to  
564 exchange with the underlying seawater (i.e. when the  $V_b$  is  $> 5\%$  after Golden et al., 1998),  
565 the saturation factor remains low and gas species remain dissolved in the brine and can be  
566 rejected to the underlying water by convection from the permeable columnar layer. This limits  
567 bubble formation, and hence the air volume fraction was  $< 1\%$  (Fig.13a and b, blue circles).  
568 Although the air volume fraction is low in these layers, it is somewhat surprising that the air  
569 volume fraction is  $> 0$ ; in theory, bubble nucleation occurs when  $SAT_f > 1$ , so these  
570 subsaturated layers should be bubble-free, though bubble nucleation from saturated gas  
571 solutions has been observed at much lower saturations than theoretically expected (Lubetkin,  
572 2003). On January 14, 75% of the bubbles observed were located in subsaturated permeable  
573 bottom layer of columnar sea ice. On January 16 and 25, 11% and 13% (respectively) of the  
574 air inclusions observed were situated in subsaturated permeable sea ice. Bubble nucleation  
575 processes are favoured where (i) there are geometrical imperfections (Wilt 1986); (ii) at  
576 “active sites” on a heterogeneous surface that can be chemically, structurally, or geometrically  
577 inhomogeneous (Deutscher and Fletcher, 1990; Kozisek et al. 2000); and (iii) by  
578 heterogeneous supersaturation away from thermodynamic equilibrium (Li and Yortsos 1994),  
579 conditions which are all met in sea ice. The contact of the three-phase (solid ice, liquid brine,  
580 air bubble and precipitated salt) in brine inclusions produced a highly heterogeneous surface,  
581 which is both chemically and structurally inhomogeneous. Moreover, full-depth convection  
582 on January 14 and convection confined to the permeable subsaturated bottom columnar layer  
583 of sea ice ( $V_b \approx 20\%$ ) on 16 and 25 January likely produced local fluctuations in the amount  
584 of gas-saturated liquid, creating the possibility of local or episodic supersaturation that may

585 have produced bubbles as has been observed by Zhou et al., (2013). Convection driven  
586 nucleation processes likely produced micro bubbles in columnar permeable sea ice, which  
587 contributed to 8.4% of the total air volume fraction observed (Table 6.). Therefore, brine  
588 drainage is only effective for the transport of dissolved gases to the underlying seawater. The  
589 rejection of dissolved gas contributes to maintain gas concentrations close to the equilibrium.  
590 Nucleation processes driven by the convective exchange in the bottom layer however limited  
591 by the saturation state increase the total gas content ( $\text{ml L}^{-1}$  ice) of sea ice by ensuring that gas  
592 trapped in bubbles remains within the sea ice and is not rejected to the underlying water  
593 (Tison et al., 2002).

594 On January 16 and 25, we observed a slight increase of air volume fraction at the  
595 transition between the subsaturated and permeable columnar sea ice and the supersaturated  
596 impermeable columnar sea ice at two-thirds of the total sea ice thickness (isotherm  $-4.1^{\circ}\text{C}$  and  
597  $-3.8^{\circ}\text{C}$ , respectively) (Fig. 7b, shaded pink area). This imparts that bubbles created by  
598 convection-driven nucleation in the permeable bottom layer, naturally accumulate at the brine  
599 permeability transition as result of their buoyancy, trapped when the sea ice matrix becomes  
600 impermeable to liquid. Our work indicates that brine will start to supersaturate ( $\text{SAT}_f = 2.7$  to  
601  $5$ ) under cooling (isotherm  $-4.1^{\circ}\text{C}$  and  $-3.8^{\circ}\text{C}$ , respectively) when the sea ice begins to  
602 become vertically impermeable to liquid, leading to solubility-driven nucleation. During ice  
603 growth period, we could expect an increase of air volume fraction above the permeable  
604 bottom layer forming a layer of entrapped bubbles. As long as the intermediate columnar ice  
605 stays impermeable (i.e. absence of warming), this bubbly transition layer will grow thicker as  
606 the ice thickens.

#### 607 **4.3.2 Intermediate impermeable columnar sea ice**

608 Within the supersaturated impermeable columnar layer, bubble nucleation is solubility  
609 driven and we expect the air volume fraction to be a function of the saturation factor. Within  
610 the supersaturated impermeable columnar layer, the air volume fraction becomes increasingly  
611 a function of the saturation factor as the brine volume increases (Fig. 13b, red circles). At low  
612 brine volumes, the air volume fraction is low regardless of the saturation factor, as indicated  
613 by the accumulation of red circles in the top left corner of Fig. 13b. As brine volume increases  
614 in the impermeable supersaturated intermediate columnar layer, both air volume fraction and  
615 bubble size increase (Fig. 13a and b). At a given  $\text{SAT}_f$ , there are more gas molecules available  
616 to go into the gas phase when brine volume increases, thereby increasing the air volume  
617 fraction and the size of existing bubbles. We therefore suggest that bubble nucleation is a

618 function of the saturation factor as well as the brine volume. Solubility-driven nucleation  
619 produced micro bubbles and large bubbles depending on the brine volume in this layer,  
620 contributed 44% of the total observed air volume fraction (Table 6).

### 621 **4.3.3 Granular sea ice**

622 We observed an increase of air volume fraction nearest the ice-atmosphere interface  
623 and generally within the ice surface granular ice layer (Figs 7d-e. and 9b). This granular  
624 surface layer had the highest gas content, the highest saturation factor, the highest air volume  
625 fraction ( $5\% < V_a < 19.4\%$ ) and contained bubbles with diameters as large as 4.5, 13, and 18  
626 mm on January 14, 16 and 25, respectively (Table 5). The increase of air volume fraction and  
627 the total gas content (largely underestimated) in the surface granular layer can be explained by  
628 a combination of several processes.

629 The formation of frazil ice is well known to contain more gas than columnar ice  
630 because it traps gas directly from the atmosphere (Tsurikov, 1979; Cole et al., 2004, Zhou et  
631 al., 2013) can explain the air volume fraction in the sea ice formed on January 14. Snow-ice  
632 formation observed thereafter on 16 and 25 January trapped gas initially contained within the  
633 snow. Moreover, rapid freezing of slush forces gas out of solution, building up the air volume  
634 fraction nearest the ice-atmosphere interface.

635 Macro bubbles are exclusively found in granular layer. They seems resulting of  
636 aggregation of discrete bubble like an aggregation of soap bubbles A succession of 0.6 mm  
637 thick transversal slices at 2.46 cm depth from January 25 is shown in Fig.14. In the first slice  
638 at +2.28 cm depth (Fig. 14, far left panel) four individual bubble bases are identifiable from  
639 which a single top bubble is formed at +2.46 cm depth (Fig.14 far right panel). The rapid  
640 freezing of slush in porous snow could potentially produce bubble aggregation. Granular sea  
641 ice and snow ice accounted for 26% of the bubble population observed, and snow ice  
642 formation accounted for 47% of the total porosity recorded indicating that physical processes  
643 associated with snow on new and young sea ice play an important role in the gas dynamics of  
644 new and young sea ice.

## 645 **4.4 The fate of gas versus the fate of salt**

646 Bulk salinity and bulk ice total gas content ( $\text{ml L}^{-1}$  ice) of sea ice is lower than in the  
647 seawater from which it formed, because gases in seawater are preferentially expelled from  
648 growing ice, along with salts (Cox and Weeks, 1983, 1988; Killawee et al., 1998; Tison et al.,  
649 2002, Loose et al., 2009, 2011). The range of total gas content values for our samples was 1.6  
650 to  $6.5 \text{ mL L}^{-1}$ , which is in the lower end of ranges reported by Matsuo and Miyake (1966),

651 Tison et al., (2002) and Crabeck et al., (2014b). Zhou et al., (2013) suggested that gas  
652 transport through sea ice occurs via processes diverging from those controlling the transport  
653 of salt. Since we do not observe similar profiles of these two parameters over time, this also  
654 suggests that the same processes do not regulate bulk ice salinity and bulk ice gas content.  
655 Rapid desalination occurred between January 14 and January 16, and the bulk salinity  
656 profile evolved towards a C-shaped profile over time (Fig. 5). In contrast, we observed a  
657 linear increase of gas content ( $\text{ml L}^{-1}$ ) (Fig. 6) and air volume fraction (Fig. 7) from the base  
658 to the sea ice surface and within sea ice as it thickened over time. The transport of gases  
659 through sea ice is different from that of the solutes because gases may be present in the form  
660 of bubbles, on top of being dissolved in the brine. Our results indicate that a great deal of the  
661 air volume fraction of sea ice exists in bubbles and not in the dissolved phase in brine,  
662 suggesting that desalination processes have a limited effect on sea ice gas content. Salts  
663 dissolved in brine can diffuse and/or be rejected in the underlying seawater during brine  
664 convection events, while bubbles are trapped in the ice matrix and can only migrate upward  
665 by buoyancy.

666 The presence of large bubbles and air volume fraction  $> 5\%$  in the top of the ice cover  
667 should potentially mediate gas fluxes over sea ice. Partitioning between gaseous phase and  
668 dissolved phase is of paramount importance for gas transport in sea ice, as it control the  
669 direction of transport – upward versus downward as well as the magnitude. If the gases are in  
670 the dissolved phase, they will be mainly transported downwards with the brines, like the other  
671 solutes. Few exceptions are the gas diffusion within the brines network that transport gases  
672 both ways in function of the concentration gradient (Crabeck et al., 2014a), and upward brine  
673 expulsion at the ice-air atmosphere. If the gases are in the gas phase (i.e. bubbles), they are  
674 only transported upward due to bubbles buoyancy. Kotovich et al., (2015, submitted)  
675 observed that air-ice gas transfer coefficients for  $\text{CO}_2$  in young permeable artificial sea ice is 5  
676 times larger during ice growth compared to ice decay. They suggest that this difference is due  
677 to the formation and transport of bubbles during ice growth. This process appears to provide a  
678 very efficient pathway to transport gases within and out of the ice. Indeed, 1D modelling  
679 suggests that bubbles migrating upward out of the ice contribute to 80% of the  $\text{CO}_2$  fluxes  
680 from sea ice to the atmosphere during ice growth (Kotovich et al., (2015, submitted).

681

## 682 **5 Conclusions and perspectives**

683 We used computed tomography X-ray imaging to quantify air inclusion distribution in  
684 sea ice, from which we derive the air volume fraction. Air inclusions are quickly and easily  
685 identified by X-ray tomography and quantitatively analyzed using segmentation techniques.  
686 The threshold selection is a crucial step requiring careful examination to provide successful  
687 results. The results from the CT X-ray analysis showed similar trends to conventional density  
688 and bulk ice total gas content ( $\text{ml L}^{-1}$  ice) measurement methods. X-ray imaging is non-  
689 destructive and allows for a direct determination of air inclusions in sea ice at high resolution  
690 with low errors and creates large datasets very quickly. However, the medical CT-scan show  
691 some limitation to resolve air inclusions in columnar sea ice, since accurate definition of air  
692 inclusion in columnar sea ice would require higher resolution. Further studies should involve  
693 Micro-Ct scan with pixel size of an order of magnitude smaller.

694 We differentiate between micro bubbles, large bubbles and macro bubbles based on  
695 their diameters. Micro bubbles are found both in the bottom columnar permeable layers ( $V_b >$   
696  $5\%$ ) and in the intermediate columnar impermeable layers ( $V_b < 5\%$ ) as well as in granular  
697 layers. Large bubbles are found more frequently where brine volume exceeded  $5\%$  and macro  
698 bubbles occur exclusively in the granular snow ice layer (i.e. ice formed by the infiltration of  
699 snow) nearest the ice-atmosphere interface. Although micro bubbles are the most abundant  
700 type of bubbles observed, they only accounted for  $14\%$  of the total air volume fraction  
701 recorded. In contrast, macro bubbles linked to granular snow ice layer accounted for  $1\%$  of  
702 the total number of bubbles but their size (volume) accounts for  $24\%$  of the total air volume  
703 fraction of the sea ice imaged. While the air volume fraction results from a mix of micro,  
704 large and macro bubbles, the factor controlling the air volume fraction is most likely the size  
705 of the air inclusions (i.e. bubbles) (Table 4).

706 We suggest that bubbles observed in the bottom subsaturated permeable layers are  
707 formed by convection-driven nucleation. Here the amount and size of the bubbles are limited  
708 by the low saturation state of the brine. Bubbles observed in impermeable columnar  
709 supersaturated sea ice are formed by solubility-driven nucleation, where the amount and  
710 bubble size is limited by the amount of brine. In growing sea ice, a maximum exists at a given  
711 depth just above the permeability transition, confirming the important role of this transition  
712 zone in shaping the vertical air volume fraction distribution. Macro bubbles located in the  
713 near-surface sea ice are linked to the presence of granular ice and the formation of snow ice  
714 (Table 6).

715 We conclude that processes regulating the vertical distribution of salts do not control  
716 the vertical distribution of gases, because most of the total gas content ( $\text{ml L}^{-1}$  ice) exists as  
717 bubbles rather than being dissolved in the brine as previously argued (Tison et al., 2002; Zhou  
718 et al., 2013, Moreau et al., 2014, Crabeck et 2014a, b). Once micro and/or macro bubbles  
719 form they are segregated from the transport pathway of dissolved salts, because bubbles will  
720 not drain out of the ice by convection due to their low density, so nucleation leads to an  
721 accumulation of gas in sea ice. Our work provides the first detailed visual demonstration and  
722 quantification of these processes.

723 As a result of the presence of large bubbles and higher air volume fraction  
724 measurements in sea ice we introduce new perspectives on processes regulating gas exchange  
725 at the ice-atmosphere interface, and note that further work should investigate, the effect of air  
726 volume fraction on sea ice permeability parameterizations. CT-X-ray imaging may allow for  
727 visualizations of transport pathways, for example the upward migration of bubbles. CT-X-ray  
728 imaging could be used to investigate the effect of different thermal and crystal texture regimes  
729 on bubble formation, dimensions, and their vertical and horizontal distribution in a large  
730 number of replicate cores from the same ice cover. This information is vital to the  
731 improvement of models involving transport of biochemical compounds and gas transfer  
732 between the ocean and the atmosphere in polar oceans.

733

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744 **References**

- 745 Bennington, K. O.: Desalination features in natural sea ice, *Journal of Glaciology*, 6(48): 845-  
746 857,1967.
- 747
- 748 Bock, C., and Eicken, H.: A magnetic resonance study of temperature-dependent  
749 microstructural evolution and self-diffusion of water in Arctic first-year sea ice, *Annals of*  
750 *Glaciology*, 40: 179 – 184, 2005.
- 751
- 752 Boespflug, X., Ross, N., Long, B. and Dumais, J.F.: Axial tomodensitometry: correlation  
753 between tomographic intensity and density of materials, *Can. J. Earth Sci.*, 31, 426–434,  
754 1994.
- 755
- 756 Carte, A. E.:Air bubbles in ice, *Proceedings of the Physical Society*, 77(495), 757-768, 1961.
- 757
- 758 Cole, D. M. and Shapiro, L. H.: Observations of brine drainage networks and microstructure  
759 of first-year sea ice, *Journal of Geophysical Research*, 103(C10): 21739-21750,1998.
- 760
- 761 Cox, G. F. N. and Weeks, W. F.: Brine drainage and initial salt entrapment in sodium chloride  
762 ice, *Cold Regions Research and Engineering Laboratory Research Report 345*. Hanover, NH,  
763 1975.
- 764
- 765 Cox, G. F. N., and Weeks W. F.: Equations for determining the gas and brine volumes in sea-  
766 ice samples, *Journal of Glaciology*, 29(102), 306 – 316, 1983.
- 767
- 768 Cox, G.F.N., Weeks, W.F.: Changes in the salinity and porosity of sea-ice samples during  
769 shipping and storage. *J. Glaciol.* 32 (112), 371–375, 1986.
- 770
- 771 Crabeck, O., Delille, B., Else, B., Thomas, D. N., Geilfus, N. X., Rysgaard, S., and  
772 Tison, J. L.: First “in situ” determination of gas transport coefficients (DO<sub>2</sub>, DAr, and DN<sub>2</sub>)  
773 from bulk gas concentration measurements (O<sub>2</sub>, N<sub>2</sub>, Ar) in natural sea ice, *J. Geophys. Res.-*  
774 *Oceans*, 119, 6655–6668, doi:10.1002/2014JC009849, 2014a.
- 775

776 Crabeck, O., Delille, B., Thomas, D. N., Geilfus, N. X., Rysgaard, S., and Tison, J. L.: CO2  
777 and CH4 in sea ice from a subarctic fjord, *Biogeosciences Discuss.*, 11, 4047-4083,  
778 doi:10.5194/bgd-11-4047-2014, 2014b.

779

780 Delille B., Vancoppenolle M., Geilfus N.-X., Tilbrook B., Lannuzel D., Schoemann V.,  
781 Becquevort S., Carnat G., Delille D., Lancelot C., Chou L., Dieckmann G.S. and Tison J.-L. :  
782 Southern Ocean CO2 sink: The contribution of the sea ice, *Journal of Geophysical Research:*  
783 *Oceans*, 119, 6340–6355 doi:10.1002/2014JC009941, 2014.

784

785 Duliu, O.G.: Computer axial tomography in geosciences: An overview. *Earth-Science*  
786 *Reviews*,48, 265-281,1999.

787

788 Garcia, H. E., and Gordon, L. I.: Oxygen solubility in seawater - better fitting equations,  
789 *Limnology and Oceanography*, 37(6), 1307-1312,1992.

790

791 Eicken, H., Bock, C., Wittig, R., Miller, H., and Poertner, H.-O.: Magnetic resonance imaging  
792 of sea-ice pore fluids: methods and thermal evolution of pore microstructure, *Cold Regions*  
793 *Science and Technology*, 31, 207-225, 2000.

794

795 Eide, L. I. and Martin, S.: The formation of brine drainage features in young sea ice, *Journal*  
796 *of Glaciology*, 14(70): 137-154, 1975.

797

798 Feltham, D. L., Untersteiner, N., Wettlaufer, J. S. and Worster, M. G.: Sea ice is a mushy  
799 layer, *Geophysical Research Letters*, 33, L14501, doi: 10.1029/2006GL026290, 2006.

800

801 Fritsen, C. H., Lytle, V. I., Ackley, S. F., and Sullivan, C. W.: Autumn bloom of Antarctic  
802 pack-ice algae, *Science*, 266(5186): 782-784, 1994.

803

804 Galley, R. J., Else, B. G. T., Geilfus, N.-X., Hare, A. A., Isleifson, D., Barber, D. G. and  
805 Rysgaard S.: Imaged brine inclusion in young sea ice – Shape, distribution and formation  
806 timing, *Cold Regions Science and Technology*, 111, 39-48,  
807 doi :10.1016/j.coldregions.2014.12.011, 2015a.

808

809 Galley, R. J., Else, B. G. T., Geilfus, N.-X., Hare, A. A., Babb, D., Papakyriakou, T., Barber,

810 D. G. and Rysgaard, S.: Micrometeorological control of frost flower growth and decay on  
811 young sea ice, *Arctic*, 68(1) :79-92, doi : 10.14430/arctic4457, 2015b.

812

813 Garcia, H. E., and Gordon L. I.: Oxygen solubility in seawater: Better fitting equations,  
814 *Limnology and Oceanography*, 37(6), 1307-1312,1992.

815

816 Geilfus, N.-X., Tison, J.-L., Ackley, S. F., Rysgaard, S., Miller L.A. and Delille, B.: Sea ice  
817 pCO<sub>2</sub> dynamics and air-ice CO<sub>2</sub> fluxes during the Sea Ice Mass Balance in the Antarctic  
818 (SIMBA) experiment - Bellingshausen Sea, Antarctica, *The Cryosphere*, 8, 2395-2407,  
819 doi:10.5194/tc-8-2395-2014, 2014.

820

821 Geilfus, N.-X, Galley, R.-J, Crabeck, O., Papakyriakou, T., Landy, J., Tison, J.-L.: Inorganic  
822 carbon dynamics of melt pond-covered first year sea ice in the Canadian Arctic  
823 *Biogeosciences Discussions* 11, 7485-7519, 2015.

824

825 Golden, K.M., Ackley, S.F., and Lytle, V.I.: The percolation phase transition in sea ice,  
826 *Science*, 282, 2238-2241, 1998.

827

828 Golden, K.M., Eicken, H., Heaton, A.L., Miner, J., Pringle, D.J., and Zhu, J.: Thermal  
829 evolution of permeability and microstructure in sea ice, *Geophys. Res. Lett.*, 34,  
830 L16501,doi:10.1029/2007GL030447, 2007.

831

832 Grenfell, T. C., A theoretical model of the optical properties of sea ice in the visible and near  
833 infrared, *J. Geophys. Res.*, 88, 9723–9735, 1983.

834

835 Hamme, R. C., and Emerson, S. R.: The solubility of neon, nitrogen and argon in distilled  
836 water and seawater, *Deep-Sea Research Part I-Oceanographic Research Papers*, 51(11), 1517-  
837 1528, 2004.

838

839 Hare, A. A., Wang, F., Barber, D., Geilfus, N.-X., Galley, R. J., and Rysgaard, S.: pH  
840 evolution in sea ice grown at an outdoor experimental facility, *Marine Chemistry*, 154: 46-54,  
841 doi: 10.1016/j.marchem.2013.04.007, 2013.

842

843 Hounsfield, G.N. :Computerized transverse axial scanning (tomography). Part 1: Description  
844 of system. *British Journal of Radiology*, 46, 1016-22, 1973.  
845

846 Hunter, M. W., Dykstra, R., Lim, M. H., Haskell, T. G., and Callaghan, P. T.: Using Earth's  
847 field NMR to study brine content in Antarctic sea ice: Comparison with salinity and  
848 temperature estimates, *Applied Magnetic Resonance*, 36: 1 – 8, doi: 10.1007/s00723-009-  
849 0003-9, 2009.  
850

851 Iassonov, P., Gebrenegus, T. and Tuller, M.:Segmentation of X-ray computed tomography  
852 images of porous materials: A crucial step for characterization and quantitative analysis of  
853 pore structures, *Water Resour. Res.*, 45, W09415, doi:10.1029/2009WR008087, 2009.  
854

855 Jones, S. F., Evans, G. M. and Galvin, K. P.: Bubble nucleation from gas cavities - a review,  
856 *Advances in Colloid and Interface Science*, 80, 27-50, 1999.  
857

858 Kapur, J. N., Sahoo, P. K. and Wong, A. K. C.: A new method for gray-level picture  
859 thresholding using the entropy of the histogram, *Graphical Models Image Process.*, 29, 273 –  
860 285, doi:10.1016/0734- 189X(85)90125-2, 1985.  
861

862 Kawamura, T. :Observations of the internal structure of sea ice by X ray computed  
863 tomography, *J. Geophys. Res.*, 93(C3), 2343–2350, doi:10.1029/JC093iC03p02343, 1988.  
864

865 Killawee, J.A., Fairchild, I.J., Tison, J.L., Janssens,L., and Lorrain, R.: Segregation of solutes  
866 and gases in experimental freezing of dilute solutions: Implications for natural glacial  
867 systems, *Geochimica Et Cosmochimica Acta*, 62, 3637-3655, 1998.  
868

869 Knoll GF.: *Radiation Detection and Measurment*. John Wiley &Sons: New York,1989.  
870

871 Kotovitch M., Moreau S., Zhou J., Vancoppenolle M., Dieckmann G. S., Evers K.-U., Van  
872 der Linden F., Thomas D. N., Tison J.-L. and Delille B.: Air-ice carbon pathways inferred  
873 from a sea ice tank experiment, submitted for publication to *Journal of Elementa*, 2015.  
874

875 Krembs, C., Gradinger, R., and Spindler, M.: Implications of brine channel geometry and  
876 surface area for the interaction of sympagic organisms in Arctic sea ice, *Journal of*  
877 *Experimental Marine Biology and Ecology*, 243: 55-80, 2000.

878

879 Langway, C. C.: Ice fabrics and the universal stage Rep. 62, U.S. Snow, Ice and 496  
880 Permafrost Research Establishment, Wilmette, Illinois, 1958.

881

882 Light, B., Maykut, G. A. and Grenfell, T.C.: Effects of temperature on the microstructure of  
883 first-year Arctic sea ice, *J. Geophys. Res.*, 108, 3051, doi:10.1029/2001JC000887, C2, 2003.

884

885 Loose, B., McGillis, W.R., Schlosser, P., Perovich, D., and Takahashi, T.: Effects of freezing,  
886 growth, and ice cover on gas transport processes in laboratory seawater experiments,  
887 *Geophysical Research Letters*, 36, L05603, L05603 10.1029/2008gl036318, 2009.

888

889 Loose, B., Schlosser, P., Perovich, D., Ringelberg, D., Ho, D.T., Takahashi, T., Richter-  
890 Menge, J., Reynolds, C.M., McGillis, W.R., and Tison, J.L.: Gas diffusion through columnar  
891 laboratory sea ice: implications for mixed-layer ventilation of CO<sub>2</sub> in the seasonal ice zone,  
892 *Tellus Series B-Chemical and Physical Meteorology*, 63, 23-39, 10.1111/j.1600-  
893 0889.2010.00506.x, 2011.

894

895 Lubetkin, S. D.: Why is it much easier to nucleate gas bubbles than theory predicts?  
896 *Langmuir*, 19, 2575– 258, 2003.

897

898 Moreau, S., Vancoppenolle, M., Zhou, J., Tison, J.-L., Delille, B., and Goosse, H.: Modelling  
899 argon dynamics in first-year sea ice, *Ocean Modelling*, 73(0), 1-18, 2014.

900

901 Matsuo, S., and Miyake, Y.: Gas composition in ice samples from Antarctica, *Journal of*  
902 *Geophysical Research*, 71, 5235-5241, 1966.

903

904 Nomura, D., Yoshikawa-Inoue, H., and Toyota, T.: The effect of sea-ice growth on air-sea  
905 CO<sub>2</sub> flux in a tank experiment, *Tellus Series B-Chemical and Physical Meteorology*, 58, 418-  
906 426, 2006.

907

908 Nomura, D., Yoshikawa-Inoue, H., Toyota, T. and Shirasawa. K. : Effects of snow, snow-  
909 melting and re-freezing processes on air-sea ice CO<sub>2</sub> flux, *J. Glaciol.*, 56(196), 262–270,  
910 2010.

911

912 Notz, D., and Worster. M. G.: In situ measurements of the evolution of young sea ice, *J.*  
913 *Geophys. Res.*, 113, C03001, doi:10.1029/ 2007JC004333, 2008.

914

915 Obbard, R., Troderman, G., Baker, I.: Imaging brine and air inclusions in sea ice using micro-  
916 X-ray computed tomography, correspondence, *J. Glaciol.*, 55, 1113–1115, 2009.

917

918 Otsu, N. : A threshold selection method from gray-level histograms, *IEEE Trans. Syst. Man*  
919 *Cybern.*, 9, 62 – 66, doi:10.1109/TSMC.1979.4310076, 1979.

920

921 Pal, N. R. : Minimum cross entropy threshold selection, *Pattern Recognition*, 29, 575–580,  
922 doi:10.1016/0031-3203(95)00111-5, 1996.

923

924 Pal, N. R., and S. K. Pal. : Entropic thresholding, *Signal Process.*, 16, 97–108,  
925 doi:10.1016/0165-1684(89)90090-X, 1989.

926

927 Papakyriakou, T., and Miller,L.: Springtime CO<sub>2</sub> exchange over seasonal sea ice in the  
928 Canadian Arctic Archipelago, *Annals of Glaciology*, 52(57), 215-224, 2011.

929

930 Perovich, D. K., and A. J. Gow. : A quantitative description of sea ice inclusions, *Journal of*  
931 *Geophysical Research: Oceans*, 101(C8), 18327-18343, 1996.

932

933 Pringle, D. J., Miner, J. E., Eicken, H. and Golden, K. M. : Pore space percolation in sea ice  
934 single crystals, *J. Geophys. Res.*, 114, C12017, doi:10.1029/2008JC005145, 2009.

935

936 Raynaud, D., Delmas, R., Ascencio, M., and Legrand, M.: Gas extraction from polar ice  
937 cores: a critical issue for studying the evolution of atmospheric CO<sub>2</sub> and ice-sheet surface  
938 elevation, *Annals of Glaciology*, 3, 265-268,1982.

939

940 Rees Jones D. W., and Worster, M. G.: A simple dynamical model for gravity drainage of  
941 brine from growing sea ice, *Geophysical Research Letters*, 40:1-5, doi:  
942 10.1029/2012GL054301, 2013.

943

944 Ridler, T. W., and Calvard, S.: Picture thresholding using an iterative selection method, *IEEE*  
945 *Trans. Syst. Man Cybern.*, 8, 630–632, 1978.

946

947 Rysgaard, S., Wang, F., Galley, R. J., Grimm, R., Notz, D., Lemes, M., Geilfus, N.-X.,  
948 Chaulk, A., Hare, A. A., Crabeck, O., Else, B. G. T., Campbell, K., Sørensen, L. L.,  
949 Sievers, J., Papakyriakou, T.: Temporal dynamics of ikaite in experimental sea ice, *The*  
950 *Cryosphere*, 8(4), 1469–1478, doi:10.5194/tc-8-1469, 2014.

951

952 Semiletov, I.P., Makshtas, A., Akasofu, S. I., and Andreas, E.L.: Atmospheric CO<sub>2</sub> balance:  
953 The role of Arctic sea ice, *Geophys. Res. Lett.*, 31, L05121, doi:10.1029/2003GL017996,  
954 2004.

955

956 Sezgin, M., and Sankur . B.: Survey over image thresholding techniques and quantitative  
957 performance evaluation, *J. Electron. Imaging*, 13, 146– 165, doi:10.1117/1.1631315, 2004.

958

959 Shannon, C. E., and Weaver. W.: The mathematical theory of communication, *Bell Syst.*  
960 *Tech. J.*, 27, 379–423, 1948.

961

962 Skoog, D. A., West, D. M. and Holler F. J.: *Chimie Analytique*, De Boeck Univ., Paris, 1997.

963

964 Stefels, J., G. Carnat, J. W. H. Dacey, T. Goossens, J. T. M. Elzenga, and J. L. Tison. : The  
965 analysis of dimethylsulfide and dimethylsulfoniopropionate in sea ice: Dry-crushing and  
966 melting using stable isotope additions, *Mar. Chem.*, 128–129, 34–43, 2012.

967

968 Timco, G. W., and Frederking. R. M. W.: A review of sea ice density, *Cold Regions Science*  
969 *and Technology*, 24, 1-6, 1996.

970

971 Tison, J. L., Haas, C., Gowing, M. M., Sleewaegen, S., and Bernard, A.: Tank study of  
972 physico-chemical controls on gas content and composition during growth of young sea ice,  
973 *Journal of Glaciology*, 48, 177-191, 2002.

974  
975 Tsai, D. M. :A fast thresholding selection procedure for multimodal and unimodal histograms,  
976 Pattern Recognition Lett., 16, 653 – 666, doi:10.1016/0167-8655(95)80011-H, 1995.  
977  
978 Tsurikov, V. :The formation and composition of the gas content of sea ice, J Glaciol, 22(86),  
979 67-81,1979.  
980  
981 UNESCO. : Eight report of the joint panel on oceanographic tables and standards, UNESCO  
982 Tech. Pap. Mar. Sci., 28 pp.,1978.  
983  
984 Vancoppenolle M, Meiners KM, Michel C, Bopp L, Brabant F, Carnat G, Delille B, Lannuzel  
985 D, Madec G, Moreau S, Tison J-L, van der Merwe P. : Role of sea ice in global  
986 biogeochemical cycles: Emerging views and challenges, Quaternary Science Reviews,  
987 79:207-230, doi:10.1016/j.quascirev.2013.04.011,2013.  
988  
989 Worster, M. G.:The dynamics of mushy layers, in Interactive Dynamics of Convection and  
990 Solidification, NATO ASI Ser. E219, pp. 113-138, Dordrecht : Kluwer, 1992.  
991  
992 Worster, M. G.: Convection in mushy layers, Annual Review of Fluid Mechanics, 29, 91-  
993 122,1997.  
994  
995 Yen, J.-C., Chang, F. J. and Chang, S.:A new criterion for automatic multilevel thresholding,  
996 IIEC Trans. Image Process., 4, 370 – 378, doi:10.1109/83.366472,1995.  
997  
998 Zack, G. W., Rogers, W. E. and Latt. S. A.: Automatic measurement of a sister chromatid  
999 exchange frequency, J. Histochem. Cytochem., 25, 741– 753.1997.  
1000  
1001 Zemmeling, H. J., Delille, B., Tison, J. L., Hints, E. J., Houghton, L., and Dacey, J. W. H.:  
1002 CO<sub>2</sub> deposition over the multi-year ice of the western Weddell Sea, Geophysical Research  
1003 Letters, 33,L13606, doi:10.1029/2006GL026320, 2006.  
1004  
1005 Zhou, J. Y., Delille, B., Eicken, H., Vancoppenolle, M., Brabant, F., Carnat, G., Geilfus, N.  
1006 X., Papakyriakou, T., Heinesch, B., and Tison, J. L.: Physical and biogeochemical properties

1007 in landfast sea ice (Barrow, Alaska): Insights on brine and gas dynamics across seasons,  
1008 Journal of Geophysical Research-Oceans, 118, 3172-3189, 10.1002/jgrc.20232, 2013.

1009 **Table**

Temperature (°C)	Salinity	Length (cm)	Ice cube volume (cm <sup>3</sup> )	Masse (g)	Density (g cm <sup>3</sup> )	V <sub>b</sub> (%)	V <sub>a</sub> (%)
-5.00	5.00	4.93	119.82	113.75	0.95	5.1	-1.3
-5.00	5.00	5.00	125	113.75	0.91	4.9	2
-5.00	5.00	5.07	130.32	113.75	0.87	4.7	6.3
<b>Relative standard error</b>		<b>±1.4%</b>	<b>±4.2%</b>		<b>±4.4%</b>	<b>±4.1%</b>	<b>±163%</b>

1010 **Table 1.** Effect of dimensional error on brine volume and air volume fraction computed by  
 1011 mass-volume density measurement using state equation from Cox and Weeks (1983).  
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 1013

V <sub>air</sub> =50%	V <sub>air</sub> x Hu <sub>air</sub>	0%<V <sub>ice</sub> <50%	V <sub>ice</sub> x Hu <sub>ice</sub>	0%<V <sub>brine</sub> <50%	V <sub>brine</sub> x Hu <sub>brine</sub>	Hu value*
0.5	-500	0	0	0.5	100	-400
0.5	-500	0.1	-7.4	0.4	80	-427.4
0.5	-500	0.2	-14.8	0.3	60	-454.8
0.5	-500	0.3	-22.2	0.2	40	-482.2
0.5	-500	0.4	-29.6	0.1	20	-509.6
0.5	-500	0.5	-37	0	0	-577

1014  $*Hu = (V_{air} \times Hu_{air}) + (V_{ice} \times Hu_{ice}) + (V_{brine} \times Hu_{brine})$

1015 **Table 2.** Estimation of the Hu-value of a pixel containing at least 50 % of air. Assuming the  
 1016 Hu-value of air, of ice and brine are -1000, -74 and 200, respectively.  
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Data	January 14	January 16	January 25	
Ice thickness (cm)	4	8	22	
Temperature (C°)	-4.1—1.7	-8.4—1.6	-5.2--2.1	
Salinity	11.4-25.8	2.26-10.3	1.3-12.5	
Brine volume fraction (%) ( $V_b$ , liquid porosity)	11.8-58.6	2.26-20.4	1.3-20.6	
Bulk ice Density ( $g\ cm^3$ )	0.91	0.89-0.92	0.84-0.92	
Air volume fraction (%) ( $V_a$ , CT-derived air porosity)	Granular	0.69-5.09	1.8-10	0.69-19.41
	Columnar impermeable	Na	0.13-1.89	0.42-3.01
	Columnar permeable	0.18-1.25*	0.02-0.87	0.11-0.85
Air inclusions diameter (mm)	Granular	0.097-4.53	0.097-13.31	0.97-18.2
	Columnar impermeable	Na	0.097-4.86	0.097-4.96
	Columnar permeable	0.097-1.12	0.097-1.08	0.097-1.18
Gas composition (average mixing ratio)	O <sub>2</sub> %	23.7	29.4	27.24
	Ar %	1.4	2.2	1.96
	N <sub>2</sub> %	74.9	68.4	70.8
Buk ice total gas content ( $ml\ L^{-1}$ ice, O <sub>2</sub> +Ar+N <sub>2</sub> )	1.58	2.5-4.7	2-6.5	
Gas Saturation factor (SAT <sub>f</sub> )	0.82	9.5 (top) -1.2 (bottom)	16 (top)-0.9 (bottom)	

\* 98% of the air volume fraction recorded was under 1 %.

**Table 3.** Summarizes the main sea ice characteristics and sea ice properties

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Air inclusion Classification	Abundance (% of the total Nbr of air inclusions observed)	Contribution (% of the air volume fraction produced by the air inclusion)	Location
Micro	78%	29%	Columnar and Granular
Large	20.7%	47%	Columnar and Granular (Depends most likely of $V_b$ )
Macro	1.3%	24%	Granular/snow ice

**Table 4.** Classification and properties of the air inclusions. The “abundance” is the proportion of micro, large and macro bubble on the total number of air inclusions observed (100% is the total number of inclusions in the three datasets (Jan 14 + Jan 16 + Jan 25)). The “contribution” is the percentile of the porosity produced by the micro, large and macro inclusions (100% is the total of air volume fraction observed in the three data set (Jan 14+ Jan 16 +Jan 25)).

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Data		January 14	January 16	January 25
Granular	Permeability	Permeable $V_b=11\%$	Permeable $V_b=6.2\%$	Permeable $3.9\%<V_b<12\%$
	Saturation	Na	Supersaturated $SAT_f=9.5$	Supersaturated $SAT_f=16$
	Air volume fraction	0.74%-5.09%	1.8%-10%	0.69%-19.41%
	Bubble class	Micro and large	Micro, large and macro	Micro, large and macro
Intermediate columnar	Permeability	Na	Impermeable	Impermeable
	Saturation	Na	Supersaturated	Supersaturated
	Air volume fraction	Na	$V_a<2\%$	$V_a<2\%$
	Bubble class	Na	Micro and large	Micro and large
Bottom columnar	Permeability	Permeable	Permeable	Permeable
	Saturation	Subsaturated $SAT_f\leq 1.2$	Subsaturated $SAT_f\leq 1.2$	Subsaturated $SAT_f\leq 1.2$
	Air volume fraction	$V_a<1\%$	$V_a<1\%$	$V_a<1\%$
	Bubble class	90 % Micro	90 % Micro	90% Micro

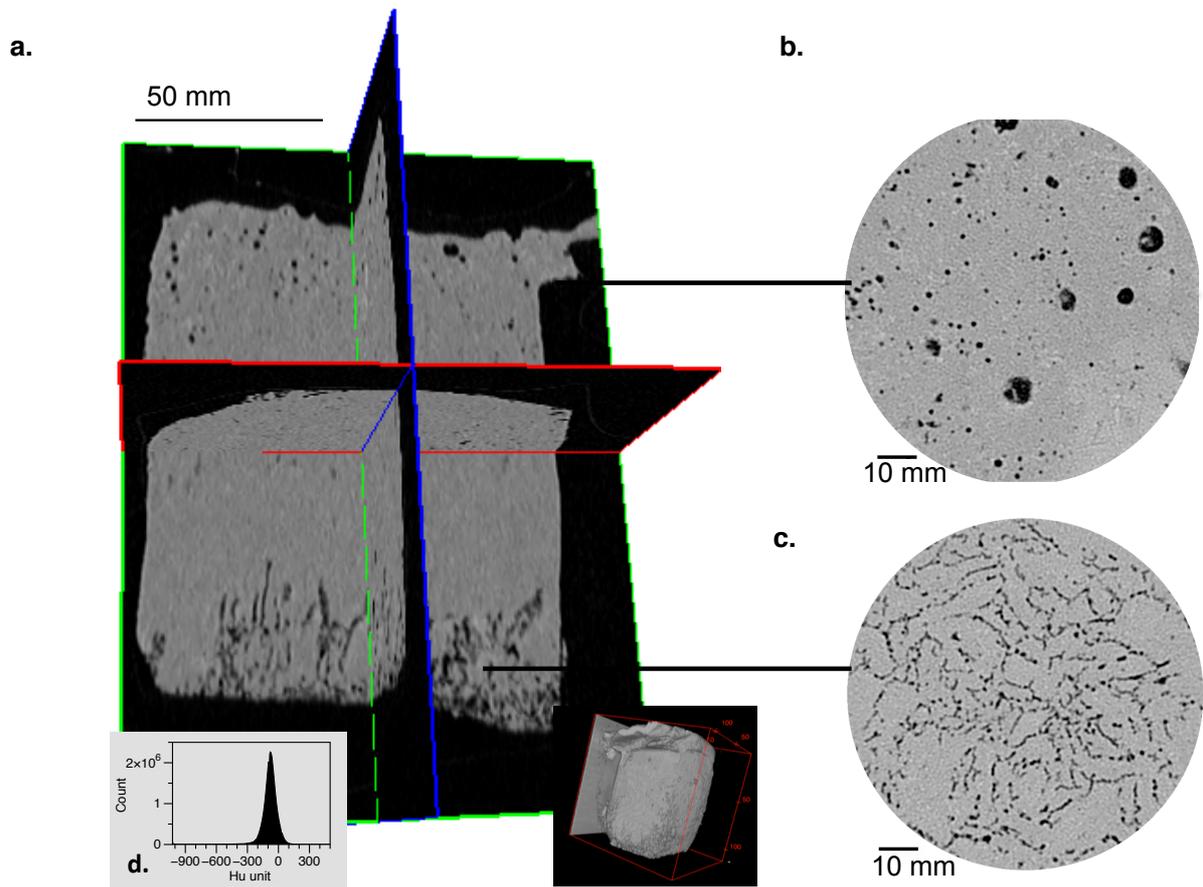
1052 **Table 5.** Physical characteristics of the various ice types. Where the brine volume exceeds the  
1053 permeability threshold for columnar ice of 5%  $V_b$  (Golden et al., 1998,2007), the ice layer is  
1054 so-called permeable.  
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Nucleation Processes	Limitation factor	Abundance (% of the total Nbr of air inclusions observed)	Contribution (% of porosity produced by the inclusions)	Type of air inclusions produced	Location
Convection driven	Saturation level	14%	8.4%	Most likely micro	subsaturated columnar
Solubility Driven	Brine volume fraction	60%	44%	Micro large	to Supersaturated Columnar
Snow ice formation		26%	47%	Micro Large Macro	Granular/ snow ice

1056 **Table 6.** Main parameters influencing the air volume fraction. The “abundance” is the  
1057 proportion of the inclusions concerned by the processes on the total of inclusion observed  
1058 (100% is the total of inclusions observed in the three data set (Jan 14+ Jan 16 +Jan 25)). The  
1059 “contribution” is the percentile of the porosity produced by the inclusions formed by  
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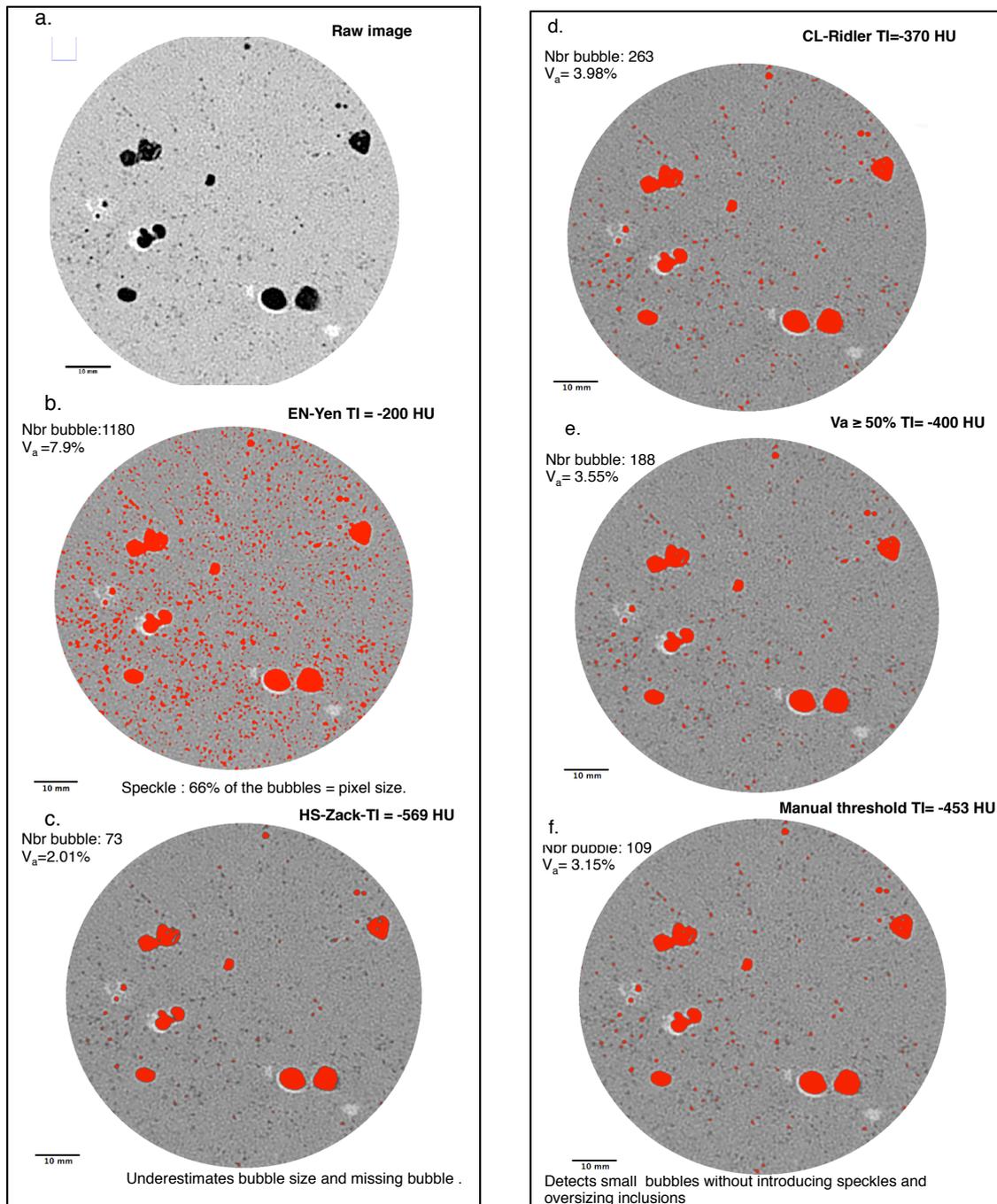
1061 convection driven, solubility driven and snow ice formation processes, respectively (100% is  
1062 the total of air volume fraction observed in the three data set (Jan 14+ Jan 16 +Jan 25)).

1063 **Fig.Captions**



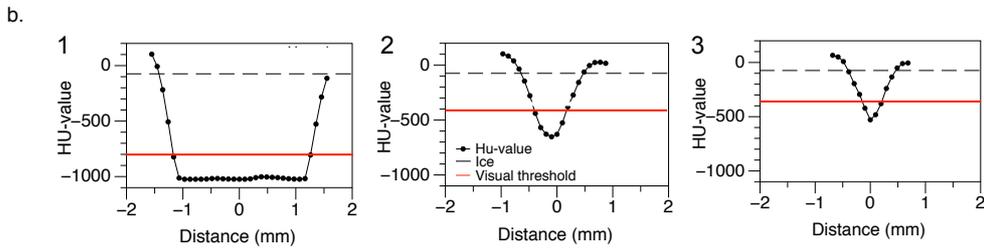
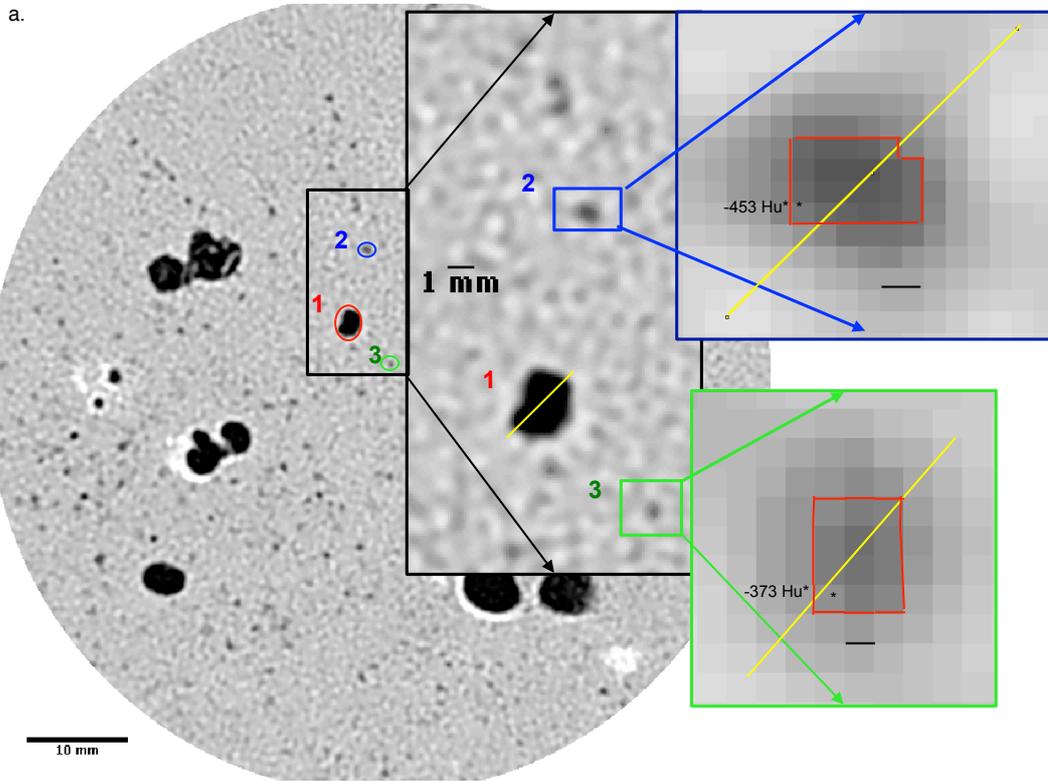
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1065 **Figure.1** (a) 3D Orthoslice view from raw images of January 16, consisting of two  
1066 longitudinal slices and one transversal slice, lighter grey represent the ice matrices and black  
1067 area represent the air inclusions (i.e bubbles); (b) Top Transversal slice at 0.65 mm depth,  
1068 every black dot represents an air inclusion (i.e bubbles); (c) Bottom transversal slice at 8 cm  
1069 depth, all the black dots show drained brine ; panel (d) Histogram of HU-unit (Ct-value )  
1070 recorded for 186 transversal slices of 0.6 mm thick for January 16.



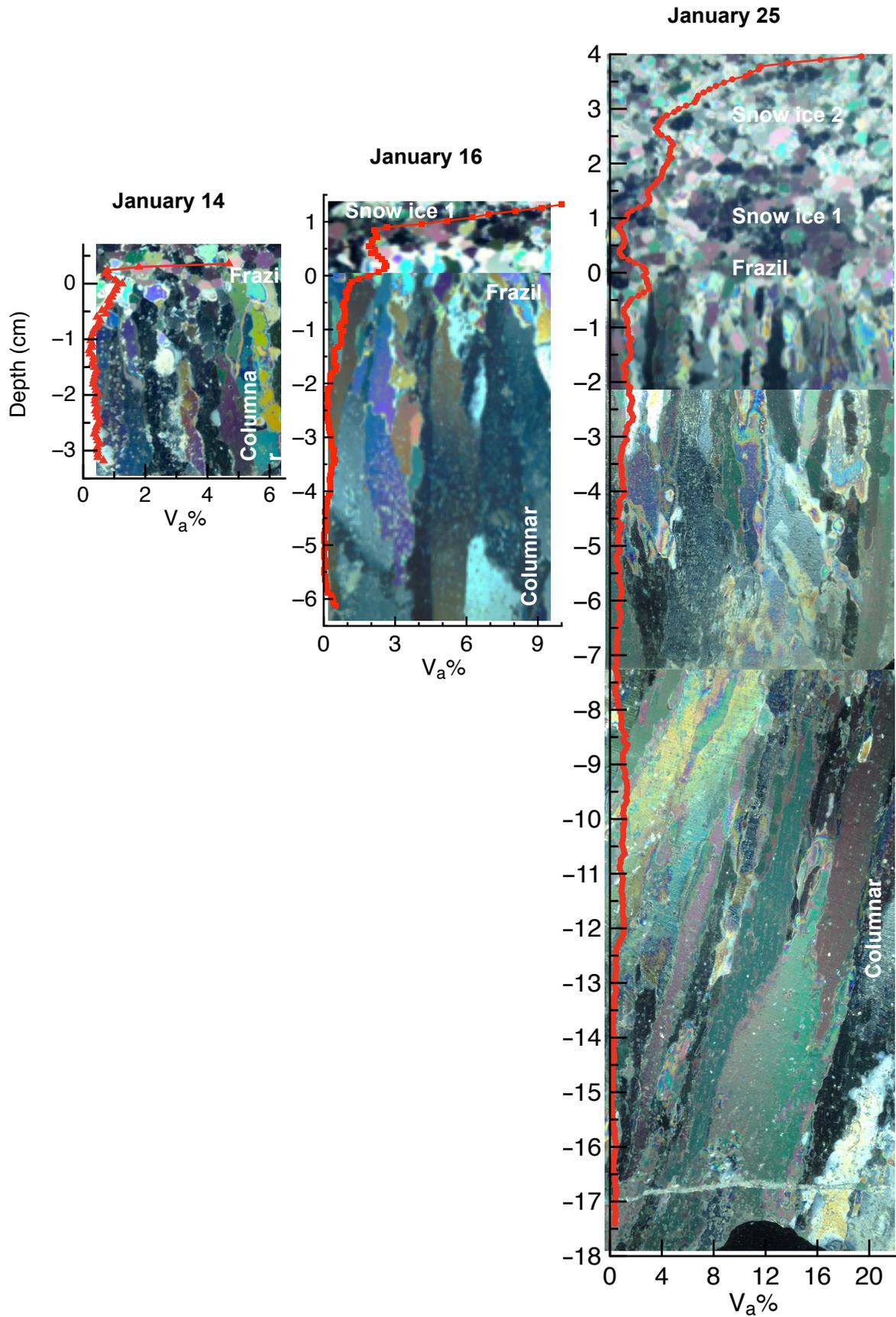
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**Figure 2.** (a) Raw transversal slice where grey pixels represent ice, black pixels represent air, and darker grey pixels are pixel-containing air. (b) and (c) Transversal slices showing the air selected pixels in red using a HU-value of -200 and of -569 as threshold selected by the EN-Yen (b) and the HS-Zack algorithm (c), respectively. (d), (e) and (f) Transversal slices showing the air selected pixels in red using a HU-value of -370, of -400 and of -453 as threshold selected by (d) CL-Ridler algorithm, (e) by pixels containing a minimum of 50% of air and (f) by manual thresholding.



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**Figure 3.** (a) Raw transversal slice where grey pixels represent ice and black pixels represent air, darker grey pixels are pixels containing air and white pixels are containing brine. We highlighted three air inclusions: an air inclusion (1) larger than the spatial resolution and two inclusions (2 and 3) which are smaller than the spatial resolution, their Hu-value never reached the Hu -value of air, instead there appeared as mixed pixels and their Hu-value reflects the proportion of air, ice and brine in the pixel. The red line shows the visual boundary of the bubble where the lowest Hu-value observed for pixel (\*) is the visual threshold. (b) The distribution of HU-value along the transects (yellow lines in panel a).



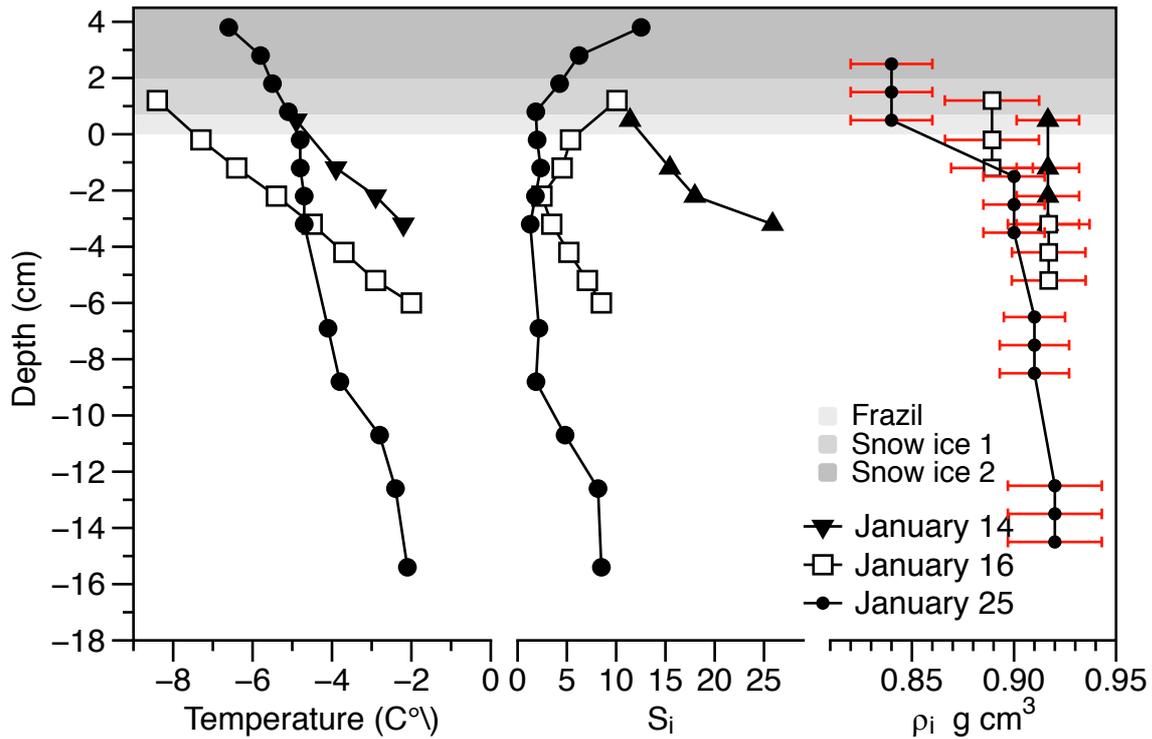
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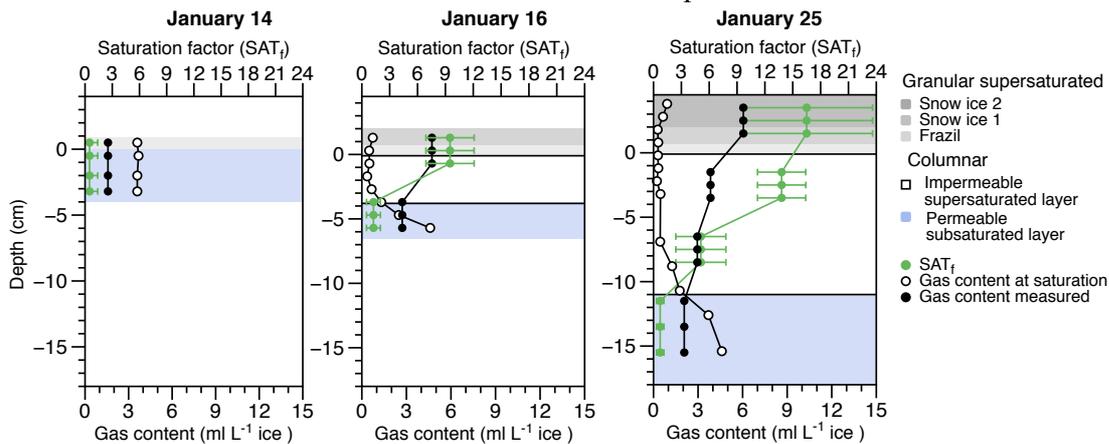
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**Figure 4.** Sea ice microstructural images overlain by the air volume fraction (red curve) for January 14, 16 and 25. The x scale differs for each date in order to visualize vertical change.

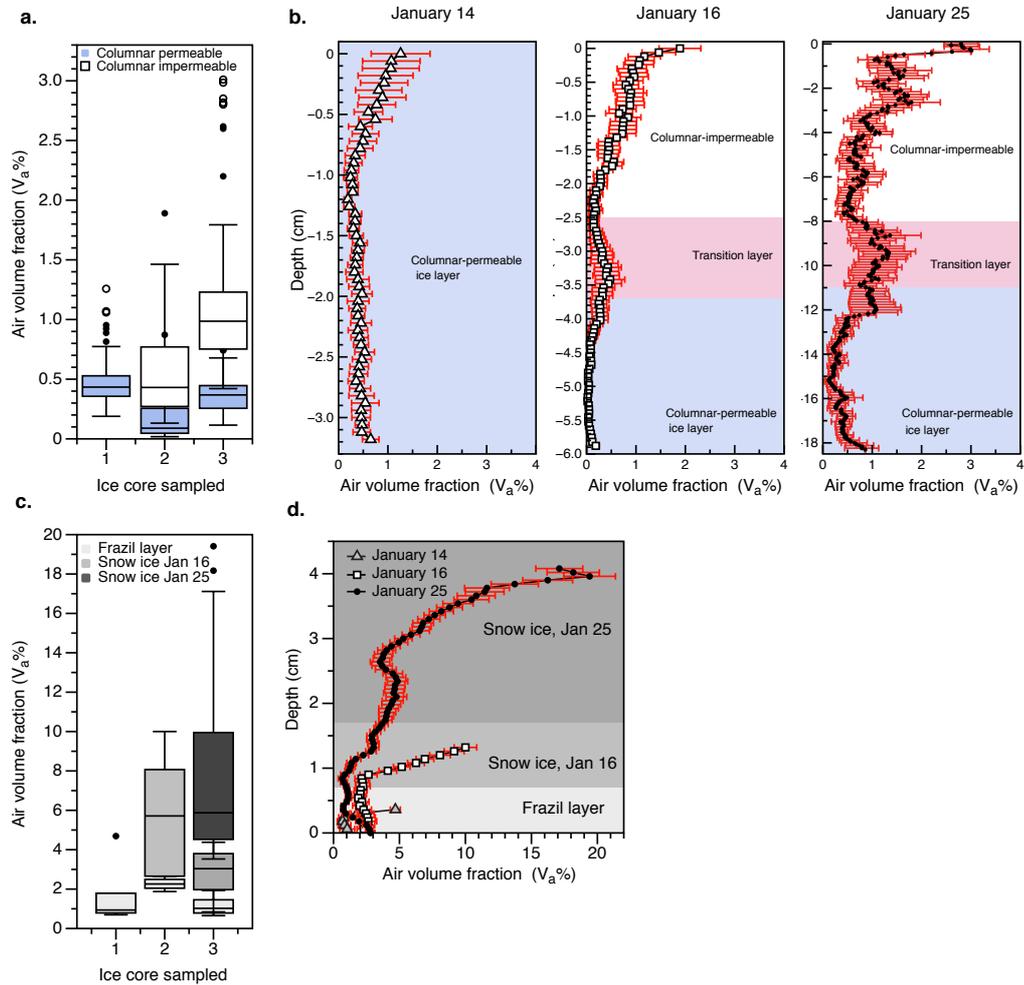
1091 The zero depth is fixed at the boundary between granular and columnar ice. Through the  
 1092 studied period ice grew from the bottom increasing the columnar layer, as well as by the top  
 1093 due to additional formation of snow ice.



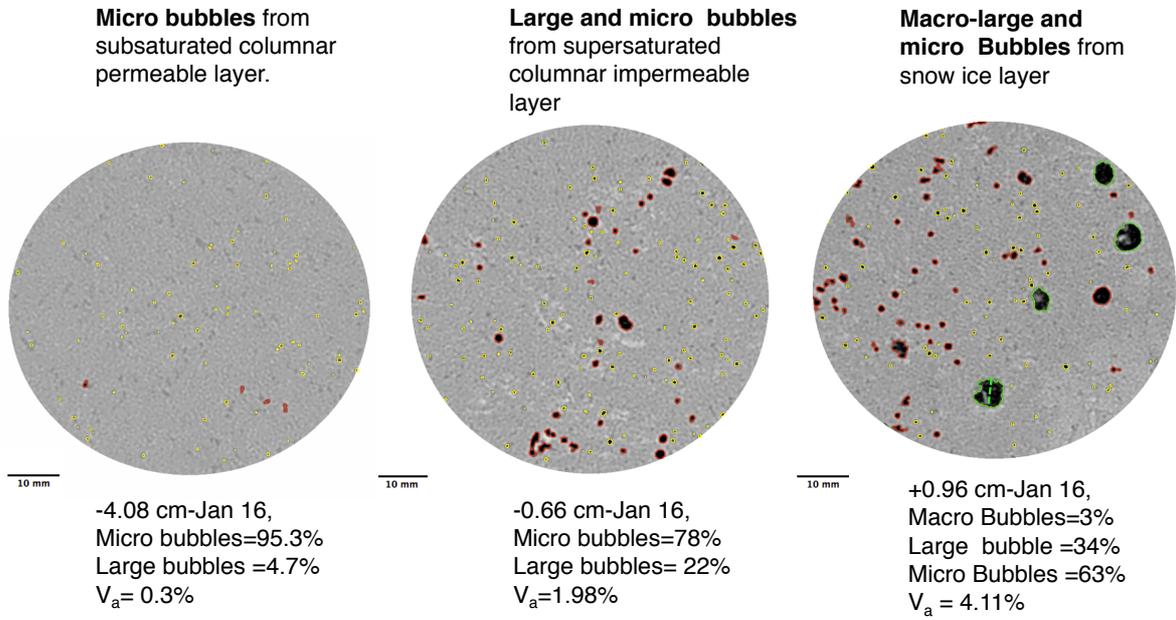
1094 **Figure 5.** Ice in situ temperature ( $^{\circ}C$ ), bulk ice salinity, and brine volume fraction ( $V_b$ ) and  
 1095 bulk ice density profiles ( $g\ cm^{-3}$ ) on 14, 16, 25 Jan. The dotted line at 5% on the  $V_b$  panel  
 1096 refers to the theoretical liquid permeability threshold for columnar sea ice (Golden et al.  
 1097 1998). Red bars on the density profiles indicate the standard deviation of the mean of density  
 1098 measurements derived from the mass-volume technique.  
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1100 **Figure 6.** Profiles of the total gas content in bulk sea ice measured by gas chromatography as  
 1101 the sum of  $O_2$ ,  $N_2$  and Ar (black symbols) compared (i) to the theoretical bulk ice gas content  
 1102 at atmospheric saturation (white symbols) and (ii) the saturation factor (green symbols). The  
 1103 blue area highlights subsaturated columnar sea ice ( $SAT_f \leq 1.2$ ), the white area highlights the  
 1104 supersaturated columnar sea ice ( $SAT_f > 1.2$ ) and the grey area represents successively the  
 1105 supersaturated granular layers (frazil and snow ice layers).  
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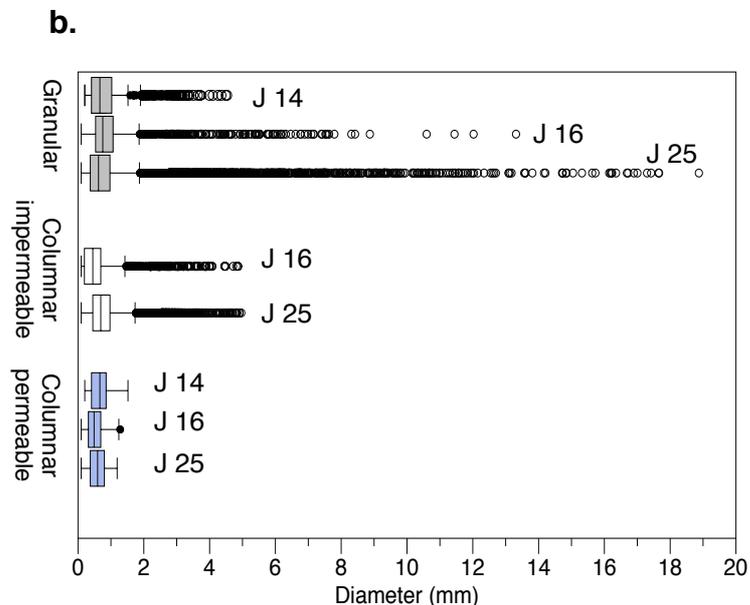
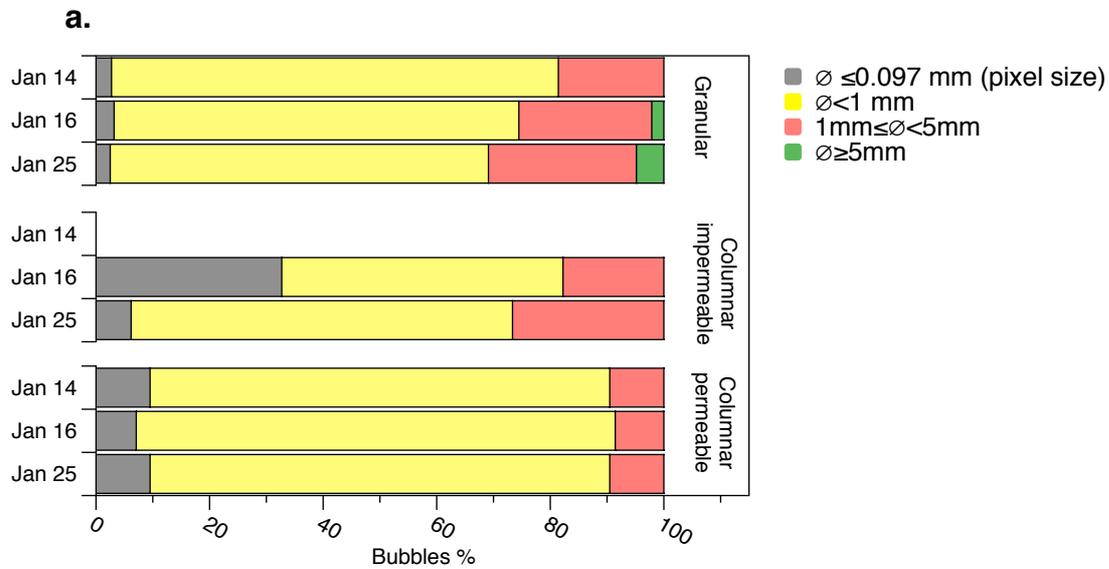


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 1110 **Figure 7.** (a) IQR box plot showing the distribution of CT-derived  $V_a$  computed for every 0.6  
 1111 mm thick slice of each ice core in the the columnar impermeable and bottom columnar  
 1112 permeable layers on January 14 (1), 16 (2) and 25 (3), respectively. The box is defined by the  
 1113 first and third quartiles of the distribution, the line in the box is the median, the circles  
 1114 represent the outliers (an outlier is any value that lies more than one and a half times the  
 1115 length of the box from either end of the box, T-bars). (b)  $V_a$  profile in the columnar layers for  
 1116 January 14, 16 and 25, respectively. The Y scale differs for each date to obtain better  
 1117 visualisation of the  $V_a$  profile. The errors bars show the potential range of CT-derived  $V_a$  in  
 1118 each transverse slice. (c) IQR box plot showing the distribution of CT-derived  $V_a$  computed  
 1119 for every 0.6 mm thick slice of each ice core in granular layers and (d)  $V_a$  profile in granular  
 1120 ice for each sampling date.  
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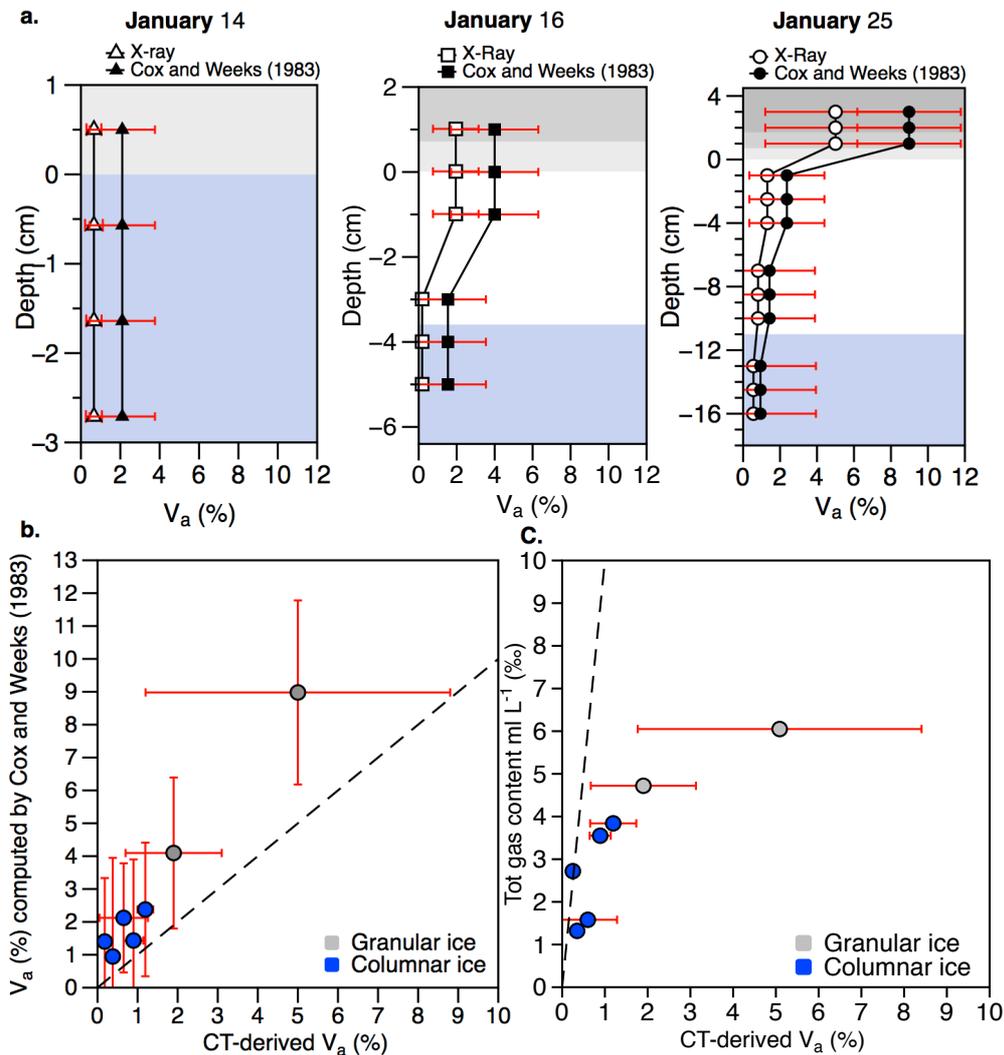


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**Figure 8.** Transversal slice at different depth highlighting the proportion of micro (yellow), large (red) and macro (green) bubble in each slice (e.g.  $[\text{Nbr micro} / (\text{Nbr micro} + \text{Nbr Large} + \text{Nbr Macro}) ] \times 100$  ).

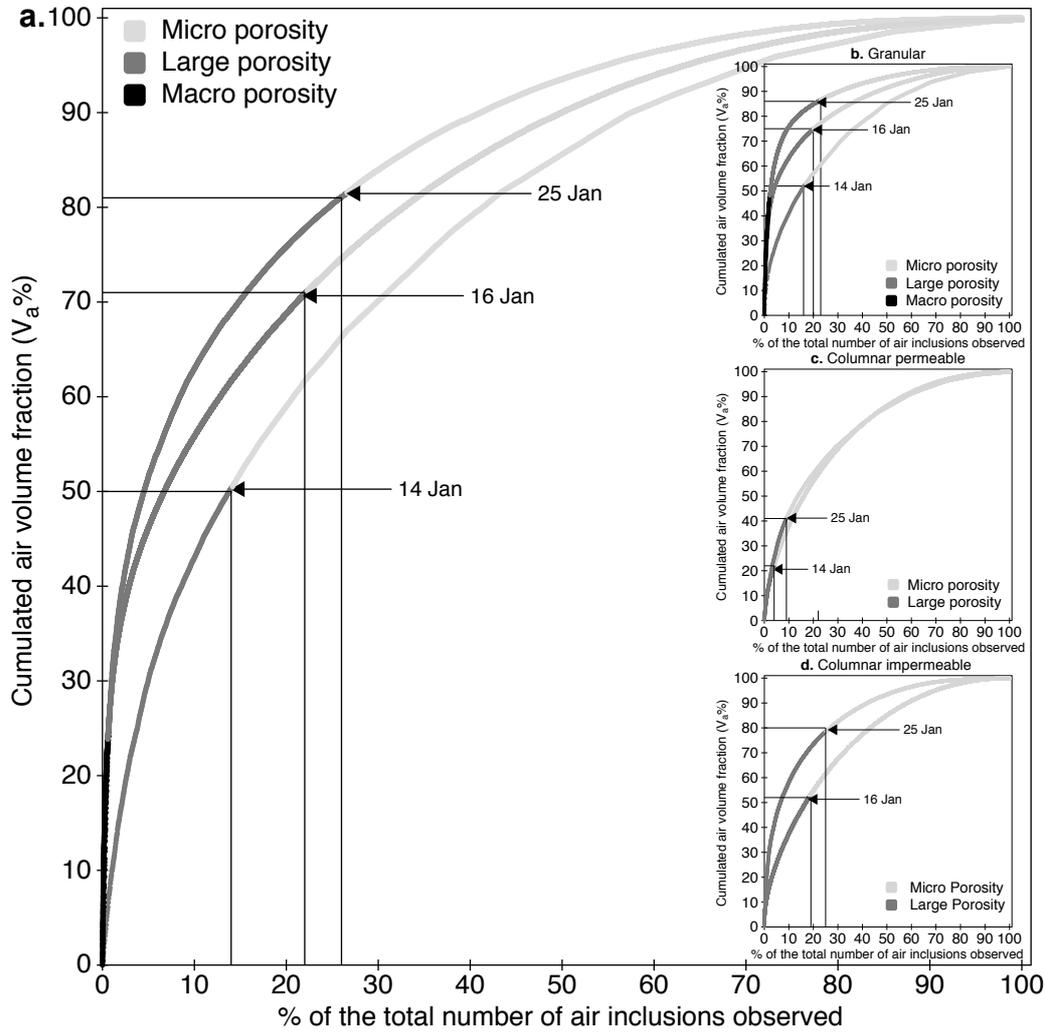


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 1129 **Figure 9. (a)** The proportion of micro, large and macro bubbles for each ice type and  
 1130 sampling date. **(b)** IQR box plot showing the distribution of the bubble diameters, per ice type  
 1131 and sampling dates. The box is defined by the first and third quartiles of the distribution, the  
 1132 line in the box is the median, the circles represent the outliers (an outlier is any value that  
 1133 lies more than one and a half times the length of the box from either end of the box, T-bars)



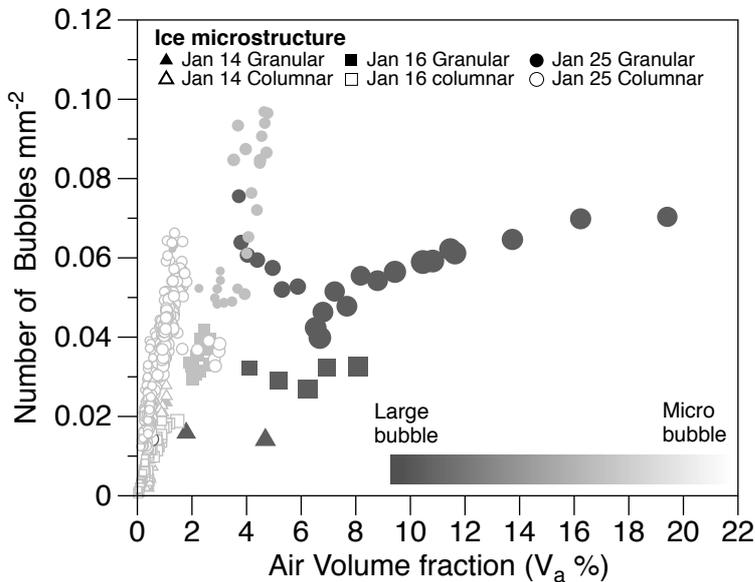
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1135 **Figure 10. (a)** Air volume fraction profiles derived from the Cox and Weeks (1983) equations  
 1136 (filled black symbols) using density measurements, with error bars showing the standard  
 1137 deviation from the mean of the results. These are compared to the CT-derived air volume  
 1138 fraction averaged for 5 cm section (filled white symbols) which error bars show the standard  
 1139 deviation of the mean along the 5 cm section. **(b)** The relationship between CT-derived air  
 1140 volume fraction and the Cox and Weeks (1983) air volume fraction where the dotted line  
 1141 signifies the 1:1 relationship. **(c)** The relationship between CT-derived air volume and the  
 1142 GC-derived bulk ice total gas content measured where the dotted line signifies the 1:1  
 1143 relationship. In (b) and (c) the CT-derived air volume fractions are averaged for 5 cm section  
 1144 and error bars show the standard deviation of the mean along the 5 cm section. Where CT-  
 1145 derived air volume fraction spans a large range of values along the section (e.g. granular ice),  
 1146 the standard deviation of CT-derived air volume fraction increases.  
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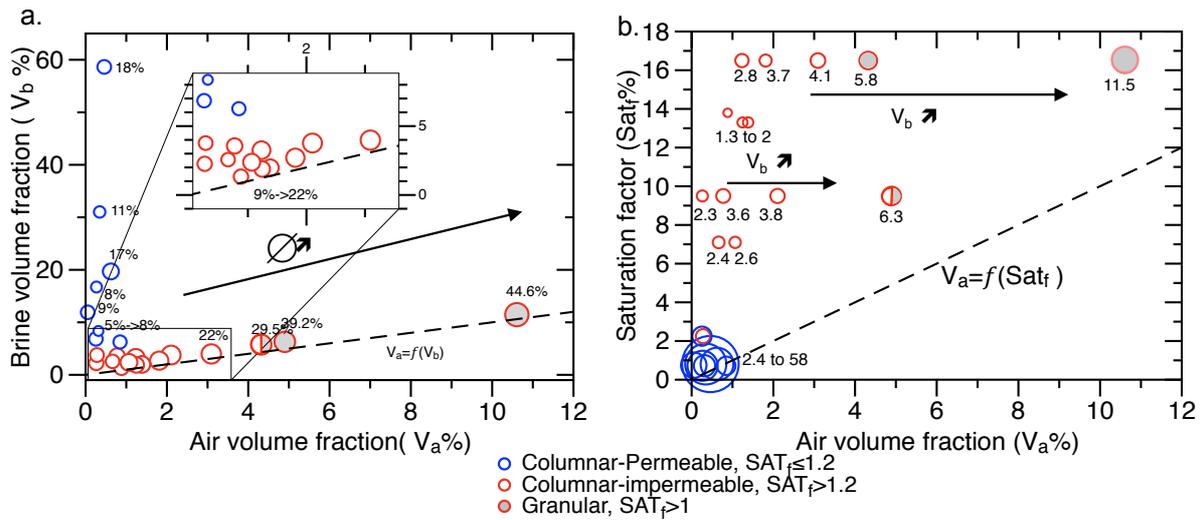
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**Figure 11.** The cumulated contribution of the macro, large and micro bubbles to the cumulated relative air volume fraction for the whole ice core (a), and in granular (b), columnar impermeable (c) and columnar permeable ice (d). It shows that, a smaller number of large bubbles (e.g. Large on 25 Jan) produced most of the air volume fraction (i.e porosity), and this is true for both the whole ice core (a) and for each type of ice (b, c, d).

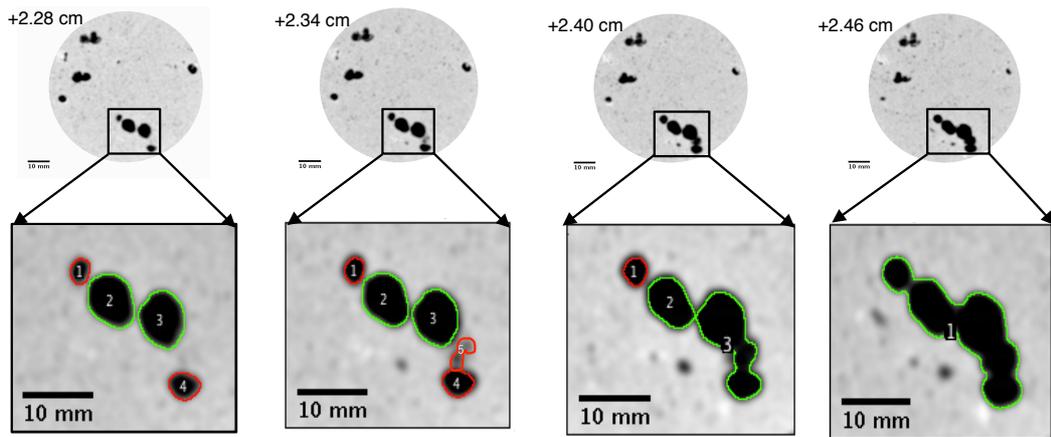


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1155 **Figure 12.** The relationship between bubble density: number of bubbles per mm<sup>2</sup> (slice area)  
 1156 and air volume fraction per slice as a function of both bubble size class and ice type (granular  
 1157 and columnar crystal texture).  
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1159 **Figure 13. (a)** Relationship between brine volume fraction and air volume fraction as a  
 1160 function of the bubble size (where the size of each marker circle is proportional to the  
 1161 percentage of bubbles with diameters  $> 1$  mm, written as % value next to some of the  
 1162 circles). **(b)** Relationship between the air volume fraction and the saturation factor as a function of the  
 1163 brine volume fraction. The size of each circle denotes the brine volume fraction (% , written  
 1164 next to some of the circles). In each panel, the bottom columnar permeable subsaturated ice (  
 1165  $SAT_f \leq 1.2$ , blue circles) is differentiated from the columnar impermeable supersaturated ice (  
 1166  $SAT_f > 1.2$ , red circles), and from the top granular supersaturated ice ( $SAT_f > 1.2$ , grey  
 1167 circles). In each panel, the dotted line signifies the 1:1 relationship.  
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1170 **Figure 14.** Four successive slices in the snow ice layer on January 25 from +2.28 to +2.46  
 1171 cm below the surface. At 2.28cm, four individual bottom end of bubbles exist where at  
 1172 2.46cm, the top end of each bubbles are joined and formed a single bubble.  
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